

## R E M A R K S

Submitted herewith is a substitute specification, excluding the claims. This substitute specification is submitted in order to introduce extensive amendments to correct minor and evident errors, and to improve the form and clarity of expression. The substitute specification submitted herewith includes no new matter.

Also submitted herewith is a replacement sheet for the sheet of drawing figures containing FIGS. 1 and 2. The changes incorporated in FIGS. 1 and 2 are submitted in order to improve the form of the legends in these drawing figures, and to conform to changes made in the specification. The replacement sheet includes no new matter. No change is being made in any drawing figures on any other sheet.

Entry of the substitute specification and replacement sheet of drawing figures is courteously requested.

Respectfully,

Christopher C. Dunham

Christopher C. Dunham  
Reg. No. 22,031  
Attorney for Applicants  
Tel. (212) 278-0400

I hereby certify that this paper is being deposited this date with the U.S. Postal Service as first class mail addressed to Commissioner for Patents, P. O. Box 1450, Alexandria, VA 22313-1450.

Christopher C. Dunham  
Christopher C. Dunham, Reg. No. 22,031

Date FEBRUARY 11, 2004

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Applicants: Y. HOTTA et al.

Serial No.: 09/877,140

Group Art Unit 1774

Filed : June 8, 2001

Examiner Bruce H. Hess

For : THERMO REVERSIBLE RECORDING MEMBER  
HAVING INFORMATION MEMORIZING PART,  
THERMO REVERSIBLE RECORDING LABEL,  
METHOD OF AND APPARATUS FOR IMAGE  
PROCESSING

**SUBSTITUTE SPECIFICATION**

**MARKED COPY**

THERMO REVERSIBLE RECORDING MEDIUM,  
MEMBER HAVING INFORMATION MEMORIZING PART,  
THERMO REVERSIBLE RECORDING LABEL, AND  
METHOD OF AND APPARATUS FOR IMAGE PROCESSING

5

FIELD OF THE INVENTION

The present invention relates to a ~~technology~~ thermo reversible recording medium, member having information memorizing part, thermo reversible recording label, and a  
10 method of and apparatus for image pressing for forming or deleting images repeatedly by utilizing a ~~heat~~ thermo sensible layer manifesting reversible change between transparent and non-transparent condition depending on temperature.

15

BACKGROUND OF THE INVENTION

Recently, attention has been given to a thermo reversible recording medium having a ~~heat~~ thermo sensible layer of which transparency can be reversed based on  
20 temperature. When such a recording medium is used in display monitors, images can be repeatedly formed or ~~deleted~~ erased when desired.

As typical examples, there are known thermo reversible recording media ~~obtained by dispersing~~ having a recording  
25 layer in which an organic lower molecular weight substance

such as a higher fatty acid ~~and the like~~ is dispersed in a matrix resin mother material such as a vinyl chloride-vinyl acetate copolymer and the like, as described. Such a medium is disclosed, for example, in Japanese Patent Application  
5 Laid-Open (JP-A) No. 55-154198.

However, in ~~ease of~~ the above-mentioned thermo reversible recording medium the transparent-state temperature range in which ~~transparency is manifested~~ the medium achieves a transparent-state is as narrow as 2 to  
10 4 °C. ~~Therefore~~ Thus, there this medium has ~~is~~ a drawback that ~~a minute temperature control, which is very difficult,~~ is required to be performed when forming or deleting images the transparent-state temperature range is too narrow to stably record and erase images.

15 ~~The~~ In attempting to widen the transparent-state temperature range ~~may be made wider by the use of,~~ higher fatty acid, higher ketone or fatty ester and an aliphatic dicarboxylic acid in admixture may be used as suggested in, for example, JP-A Nos. 2-1363, 3-2089, 4-366682 or 6-255247.  
20 Since the transparent-state temperature range becomes wider, the temperature control becomes easier.

The thermo reversible recording media are often used in, for example, point cards (a card given to a customer by a shopkeeper and in which points are added each time some  
25 service is used). Such point ~~card~~ cards are used repeatedly

over a long period of time. Such a card may be put in pant pocket, purse or in table drawer. In other words, the card is stored under various conditions or environments.

However, if basic substances such as ammonia, amine  
5 and the like exist where the card has been stored, there is a problem that an opaque image cannot be formed even if the amount of such substances is extremely small. The reason for this is supposed that a carboxyl group ~~on an~~ of the organic lower molecular weight substance ~~and a~~ reacts with the basic  
10 substance ~~forms salt which increases~~ resulting in formation of salts thereof. Therefore, the melting point of the organic lower molecular weight substance increases.

JP-A No. 5-294062 suggests use of higher ketone or fatty ester and a saturated aliphatic bisamide in admixture  
15 to make the transparent-state temperature range wider. Since organic lower molecular weight substance having a carboxyl group is not used, the influence by a the basic substance is small, the temperature is somewhat wider, ~~deleting and erasing~~ property is better. However, there  
20 is a drawback that the contrast is low.

Further, JP-A No. 11-58988 suggests, to decrease the influence by a the basic substance, use of organic lower molecular weight substance having lower melting point such as fatty ester and fatty acid metal salt such as copper  
25 stearate, or fatty amide and the like. However, if copper

stearate is used, there is a drawback that the medium becomes blue colored ~~blue~~ due to a blue color of the raw material. Moreover, if fatty amide is used, since the melting point of the fatty amide is not so higher, there is a drawback  
5 that the transparent-state temperature range becomes narrower.

#### SUMMARY OF THE INVENTION

It is an object of the present invention to provide  
10 a thermo reversible recording medium which can provide sufficient opacity even if it is stored in the presence of a basic substance. It is another object of the present invention to provide a thermo reversible recording medium which has a wider transparent-state temperature range  
15 ~~showing transparency, and by which excellent image deleting ability and high contrast image are obtained even if images~~  
can be formed or images can be erased irrespective of changes in environmental temperature changes. It is another object of the present invention to provide a thermo reversible  
20 recording label, a member having an information memorizing part and a reversible display part, an image processing method and an image processing apparatus, to which a thermo reversible recording medium, solution means of the first and second objects, is applied.

25 The thermo reversible recording medium of one aspect

of this invention comprises a substrate and a ~~heat~~ thermo sensible layer. This ~~heat~~ thermo sensible layer is made of resin and organic lower molecular weight substance and can ~~becoming~~ reversibly achieve transparent-state  
5 ~~transparent or non-transparent or vice versa~~ opaque-state depending on temperature. The organic lower molecular weight substance is a linear hydrocarbon-containing compound having no carboxyl group. The organic lower molecular weight substance may be any one or more of (1)  
10 linear hydrocarbon-containing compounds having a urethane bond, (2) linear hydrocarbon-containing compounds having a sulfonyl bond, (3) linear hydrocarbon-containing compounds having an oxalic diamide bond, (4) linear hydrocarbon-containing compounds having a diacylhydrazide  
15 bond, (5) linear hydrocarbon-containing aliphatic compounds having a urea bond and urethane bond, (6) linear hydrocarbon-containing aliphatic compounds having a urea bond and amide bond, (7) linear hydrocarbon-containing aliphatic compounds having a plurality of urea bonds, (8)  
20 linear hydrocarbon-containing cyclic compounds having a urea bond, or (9) linear hydrocarbon-containing cyclic compounds having an amide bond.

Other objects and features of this invention will become apparent from the following description with  
25 reference to the accompanying drawings.

## BRIEF DESCRIPTION OF THE DRAWINGS

Fig. 1 is a graph showing change in degree of transparency ~~by~~ of a thermo reversible recording medium ~~of~~ according to the present invention.

Fig. 2 is a graph illustrating ~~the clearing~~ an upper limit of the transparent-state temperature, specification lower limit of the opaque-state temperature, clearing initiation temperature at which the transparent-state begins and clearing temperature range of in which the transparent-state is maintained according to the thermo reversible recording medium of the present invention.

Fig. 3 is a view showing an example in which the thermo reversible recording label is pasted on a ~~MD-disk~~ cartridge of an MD disk.

Fig. 4 is a view showing an example in which the thermo reversible recording label is pasted on a CD-RW.

Fig. 5 is a view showing an example in which the thermo reversible recording label is formed on an optical information memorizing medium (CD-RW) using an AgInSbTe-based phase change-type recording material.

Fig. 6 is a view showing a display label of a video tape cassette.

Fig. 7A to Fig. 7C are views for explaining an example of the layer constitution of the thermo reversible recording



medium of the present invention.

Fig. 8 is a view illustrating one example of the use of the thermo reversible recording medium of the present invention.

5 Fig. 9A and Fig. 9B are views illustrating another example of the use of the thermo reversible recording medium of the present invention.

Fig. 10A and Fig. 10B are views illustrating still another example of the use of the thermo reversible recording  
10 medium of the present invention.

Fig. 11 is a view illustrating one example of a thermo reversible recording apparatus of the present invention.

#### DESCRIPTION OF THE PREFERRED EMBODIMENTS

15 First, the mechanism of thermo reversibility of the recording medium of the present invention will be described.

The thermo reversible recording medium of the present invention has a substrate carrying thereon a ~~heat~~ thermo sensible layer manifesting reversible change between  
20 ~~transparent condition~~ transparent-state and non-transparent ~~condition~~ (also called) or opaque-state ~~condition~~ depending on temperature, ~~and the~~ . The transparent-state condition and non-transparent condition the opaque-state are supposed to be formed according to the  
25 following mechanisms.

In the transparent-state condition, it is believed that particles of organic lower molecular weight substance dispersed in a mother material resin ~~are closely contacted~~ come in close contact with the resin ~~without~~. That is, there  
5 is no clearance or gap and ~~no clearance is present also~~ in or within the ~~partiele,~~ particles. ~~consequently~~ Consequently, light ~~entered~~ entering from one side ~~is not~~ comes out from the other side without being scattered and ~~permeates through toward the opposite side,~~ to give  
10 transparency. On the other hand, in the ~~non-transparent condition~~ opaque-state, it is believed that particles of organic lower molecular weight substance are constituted of polycrystal made of fine crystals aggregated and there is a clearance or gap ~~is formed~~ at the interface of the crystals  
15 or at the interface between the ~~partiele~~ particles and the resin ~~consequently~~. Consequently, light entering from ~~any~~ one side is refracted or reflected and scattered at the interfaces between the clearance and the ~~crystal~~ crystals and between the clearance and the resin, to give white  
20 appearance.

Fig. 1 is a view illustrating one example of ~~temperature- change in the~~ transparency ~~change~~ due to a change in the temperature of a ~~the~~ thermo reversible recording medium ~~of~~ according to the present invention.

25 In Fig. 1, a ~~heat~~ thermo sensible layer containing

a resin and an organic lower molecular weight substance dispersed in this resin as main components is, for example, in ~~non-transparent condition~~ opaque-state at room temperatures of  $T_0$  or less. When the material is heated, it begins to become transparent gradually from temperature  $T_1$ , and becomes completely transparent when the temperature is between  $T_2$  and  $T_3$ . ~~Even~~ Once the material becomes transparent it remains transparent even if the material cools to room temperature  $T_0$  or less it remains transparent.

10        The reason for this is believed ~~that~~ to be as follows. That is, the resin begins to be softened from around temperature  $T_1$ , and with the progress of softening, the resin, for example, shrinks to decrease. As a result the clearance at the interface ~~with~~ between the organic lower molecular weight substance ~~or in~~ and the particle, ~~consequently,~~ particles decreases. Thus, transparency increases gradually, and the organic lower molecular weight substance becomes semi-transparent ~~condition~~ at temperatures from  $T_2$  to  $T_3$  and the. The remaining clearance is buried with the organic

15        lower molecular weight substance to give transparency, and a . Because the material is cooled while there are seed crystal remains and is cooled to reveal crystals, crystallization begins at a relatively higher temperature ~~and the.~~ The resin is still in softened condition. ~~in this~~

20        ~~process, therefore,~~ As a result, the resin follows change

25

in volume of particles ~~by~~ due to the crystallization,  
~~consequently.~~ Thus, no clearance is formed between the  
particles and transparent-state condition is maintained.

When the resin is heated to temperature  $T_4$  or more,  
5 it becomes semi-transparent ~~between,~~ i.e. a ~~condition~~ state  
~~between the maximum total~~ transparency and the ~~maximum~~  
~~non-transparency total~~ opacity.

~~Next, when this~~ If the resin at the temperature  $T_4$   
is ~~lowered~~ cooled, the resin returns to the original  
10 ~~opaque-state non-transparent condition~~ without forming a  
transparent-state condition again.

The reason for this is believed ~~that~~ to be as follows.  
That is, after the organic lower molecular weight substance  
is completely melted at temperatures of  $T_4$  or more,  
15 super-cooling condition is formed, ~~and.~~ Therefore,  
crystallization occurs at a temperature slightly higher than  
 $T_0$ , ~~and~~ the resin can not follow the change in volume ~~by~~  
due to the crystallization, to cause generation of clearance  
and clearance is generated between particles.

20 However, a temperature-transparency change curve  
shown in Fig. 1 is only a typical example, and transparency  
at each condition may change, by variation of materials,  
depending on the material.

~~A~~ The thermo reversible recording medium forming  
25 reversibly transparent state condition and ~~non-transparent~~

~~condition opaque-state~~ depending on ~~heat~~ temperature change  
~~according to the above-described mechanism~~ as explained  
above has a basic structure ~~in which~~ as follows. That is,  
there is a substrate, and a ~~heat~~ thermo sensible layer  
5 containing a the resin and ~~an~~ the organic lower molecular  
weight substance as main components is provided on a this  
substrate.

One object of the present invention is solved by using,  
the organic lower molecular weight substance is a linear  
10 hydrocarbon-containing compound (A) having no carboxyl  
group and it may be any one of the following compounds (1)  
to (9):

- (1) linear hydrocarbon-containing compounds having  
a urethane bond,
- 15 (2) linear hydrocarbon-containing compounds having  
a sulfonyl bond,
- (3) linear hydrocarbon-containing compounds having  
an oxalic diamide bond,
- (4) linear hydrocarbon-containing compounds having  
20 a diacylhydrazide bond,
- (5) linear hydrocarbon-containing aliphatic  
compounds having a urea bond and urethane bond,
- (6) linear hydrocarbon-containing aliphatic  
compounds having a urea bond and amide bond,
- 25 (7) linear hydrocarbon-containing aliphatic

compounds having a plurality of urea bonds,

(8) linear hydrocarbon-containing cyclic compounds having a urea bond, or

(9) linear hydrocarbon-containing cyclic compounds  
5 having an amide bond.

Namely, this linear hydrocarbon-containing compound (A) ~~does~~ does not have a carboxyl group as ~~described above~~, and has a polar group as described below in the molecule.

Urethane bond (-NHCOO-), sulfonyl bond (-SO<sub>2</sub>-), oxalic  
10 diamide bond (-NHCOCONH-), diacylhydrazide bond (-CONHNHCO-), urea bond (-NHCONH-) and amide bond (-CONH-).

It is preferable that this linear hydrocarbon-containing compound (A) does not have a hydroxyl group.

15 The linear hydrocarbon-containing compound (A) may be a mixture of two or more of the compounds (1) to (9) listed above.

It is preferable that the total carbon number of linear hydrocarbons of the linear hydrocarbon-containing compound  
20 (A) is preferably between 6 and 60, more preferably between 8 and 50.

It is preferable that, among linear hydrocarbon-containing compounds (A), (1), (2), (3), (4), (8) and (9) contain a cyclic structure such as a cyclic  
25 hydrocarbon (cyclohexane, cyclopentane and the like), an

aromatic ring (benzene, naphthalene and the like), a heterocyclic ring (cyclic ether, furan, pyran, morpholine, pyrrolidine, piperidine, pyrrole, piridine, pirazine, piperazine, pyrimidine and the like), a condensed  
5 hetarocyclicring (benzopyrrolidine, indole, benzooxazine, quinoline and the like), or the like.

Further, it is preferable that the end of a molecule has a methyl group.

~~The~~ It is preferable that the melting point of the  
10 linear hydrocarbon-containing compound (A) is ~~preferably~~ above 100 °C and below 180 °C. More preferably, the melting point should be between 130 °C and 150 °C.

When the melting point is too low, the ~~clearing~~  
transparent-state temperature range cannot be made wider  
15 so that the ~~deleting ability lowers~~ image can not be erased  
effectively. On the other hand, when the melting  
~~temperature point~~ point is too high, sensitivity ~~in~~ when forming  
an opaque image decreases undesirably.

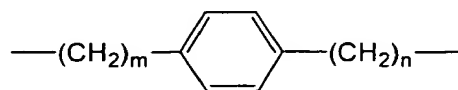
Examples of the linear hydrocarbon-containing  
20 compound (A) include, but are not limited to, compounds of  
the following general formulae (1) to (9).



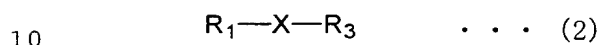
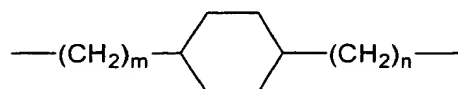
25 ~~(wherein~~ In this general formula (1), at least one of X and

Y represents a urethane bond, sulfonyl bond or urea bond, and the remaining one represents one selected from a urethane bond, sulfonyl bond, urea bond and amide bond.  $R_1$  and  $R_3$  represent  $\text{CH}_3(\text{CH}_2)_m-$  or  $\text{CH}_3(\text{CH}_2)_m-\text{O}-(\text{CH}_2)_n-$ , and  $R_2$  represents  
 5  $-(\text{CH}_2)_m-$  or a group of the following general formula (i) or (ii), and m and n are preferably from 0 to 30. ~~+~~

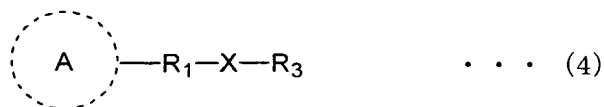
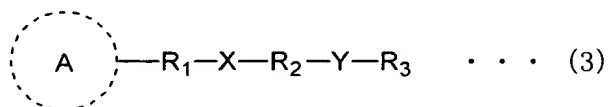
(i)



(ii)

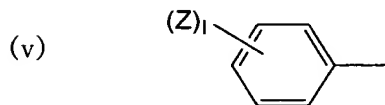
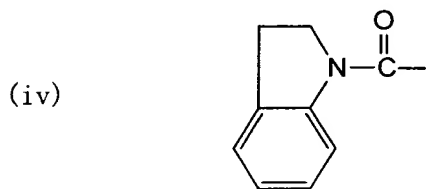
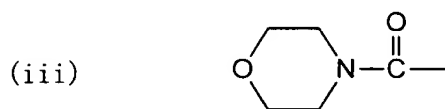


~~(wherein~~ In this general formula (2), X represents an oxalic diamide bond or diacylhydrazide bond,  $R_1$  and  $R_3$  represent  $\text{CH}_3(\text{CH}_2)_m-$  or  $\text{CH}_3(\text{CH}_2)_m-\text{O}-(\text{CH}_2)_n-$ , and m and n are preferably  
 15 from 0 to 30. ~~+~~

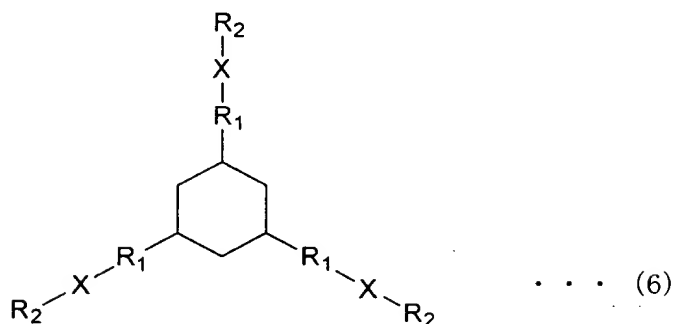
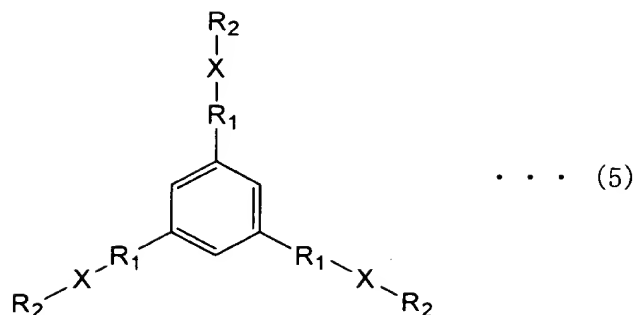




~~(in the~~ In these formulae (3) and (4), at least one of X and Y represents a urethane bond, sulfonyl bond, urea bond, amide bond, oxalic diamide bond or diacylhydrazide bond. R<sub>1</sub> and R<sub>2</sub> represent  $-(CH_2)_m-$  or  $-(CH_2)_m-O-(CH_2)_n-$ , and R<sub>3</sub> represents  $CH_3(CH_2)_m-$  or  $CH_3(CH_2)_m-O-(CH_2)_n-$ , [A] represents a phenyl group, cyclohexyl group or a group of the following general formulae (iii) to (v), and m and n are preferably from 0 to 30. ~~+-~~

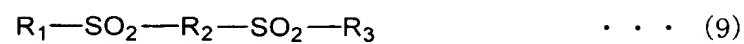
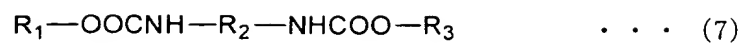


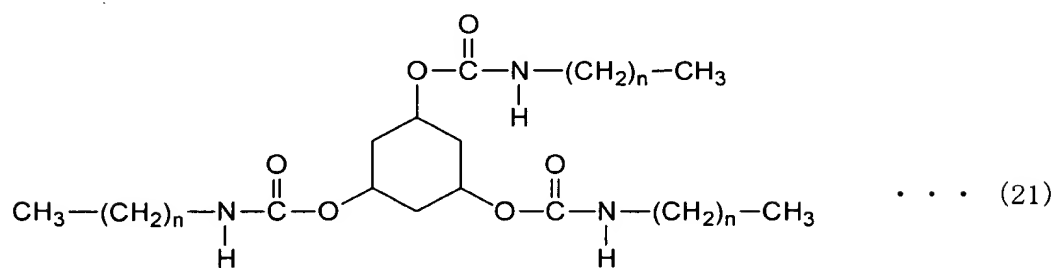
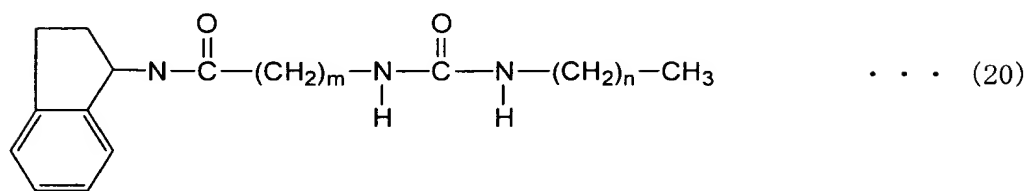
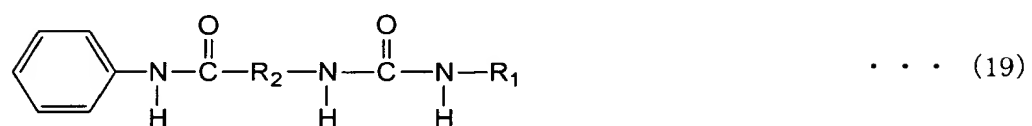
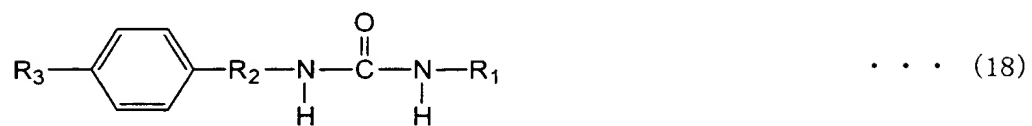
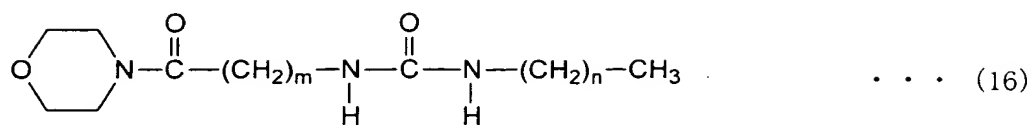
10 ~~(wherein~~ In these formulae, ~~l represents~~ l is an integer from 1 to 3 ~~+-~~ and R<sub>1</sub>OCO-, R<sub>1</sub>O-, R<sub>1</sub>, and R<sub>1</sub> represents  $CH_3(CH_2)_m-$  or  $CH_3(CH_2)_m-O-(CH_2)_n-$ , and m and n are preferably from 0 to 30. ~~+-~~

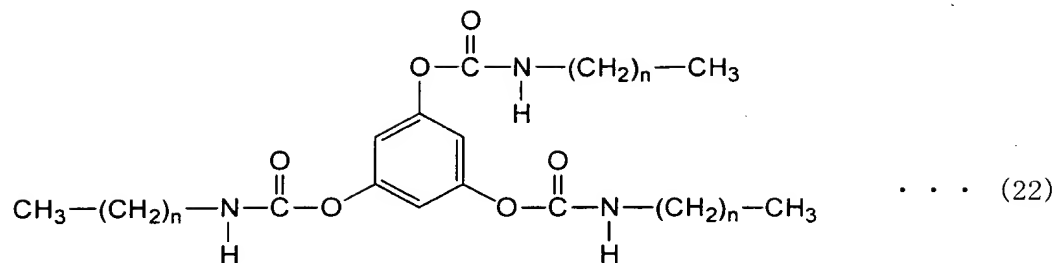


5 ~~(in the~~ In these formulae (5) and (6), X represents a urethane bond, sulfonyl bond, urea bond, amide bond, oxalic diamide bond or diacylhydrazide bond.  $R_1$  represents  $-(CH_2)_m-$  or  $-(CH_2)_m-O-(CH_2)_n-$ , and  $R_2$  represents  $CH_3(CH_2)_m-$  or  $CH_3(CH_2)_m-O-(CH_2)_n-$ , and m and n are preferably from 0 to  
 10 30. ~~+-~~

Specific examples of the linear hydrocarbon-containing compound (A) include, but not limited to, compounds ~~of~~ represented by the following general formulae (7) to (22).







~~(in the~~ In these formulae (7) to (22),  $R_1$  and  $R_3$  represent  $\text{CH}_3(\text{CH}_2)_m-$  or  $\text{CH}_3(\text{CH}_2)_m-\text{O}-(\text{CH}_2)_n-$ ,  $R_2$  represents  $-(\text{CH}_2)_m-$  or  
 5 a group of the above-mentioned general formula (i) or (ii), and  $m$  and  $n$  are preferably between 0 and 30. ~~)-~~

Wherein,  $m$  and  $n$  are preferably from 1 to 30, more preferably from 3 to 26, further preferably from 5 to 22.

The total carbon number of linear hydrocarbons in the  
 10 molecule is preferably 8 or more, more preferably 10 or more, further preferably 14 or more.

The total carbon number of linear hydrocarbons in the molecule is preferably 60 or less, more preferably 50 or less, further preferably 40 or less.

15 When the carbon number is too small, a problem occurs that compatibility with a resin is improved, a lower molecular weight particle is not formed easily, consequently, contrast decreases. On the other hand, ~~and~~ when the carbon number is too large, a problem occurs that compatibility  
 20 with a lower melting point and lower molecular weight substance lowers and the ~~clearing~~ transparent-state

temperature range can not be enlarged, undesirably widened.

As the compound of the above-mentioned general formula (7), the following compounds are exemplified.

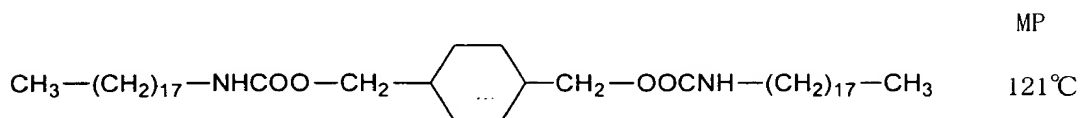
	Melting Point (MP)
$\text{CH}_3(\text{CH}_2)_{11}\text{OOCNH}(\text{CH}_2)_6\text{NHCOO}(\text{CH}_2)_{11}\text{CH}_3$	113°C
$\text{CH}_3(\text{CH}_2)_{17}\text{OOCNH}(\text{CH}_2)_6\text{NHCOO}(\text{CH}_2)_{17}\text{CH}_3$	119°C
5 $\text{CH}_3(\text{CH}_2)_{21}\text{OOCNH}(\text{CH}_2)_6\text{NHCOO}(\text{CH}_2)_{21}\text{CH}_3$	121°C

The compound of the above-mentioned general formula (8) may be any one of the following.

	MP
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_2\text{OOCNH}(\text{CH}_2)_{17}\text{CH}_3$	115°C
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_4\text{OOCNH}(\text{CH}_2)_{17}\text{CH}_3$	119°C
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_6\text{OOCNH}(\text{CH}_2)_{17}\text{CH}_3$	111°C

10

~~Chemical formula 3~~



The compound of the above-mentioned general formula (9) may be any one of the following.

15

	MP
$\text{CH}_3(\text{CH}_2)_{11}\text{SO}_2(\text{CH}_2)_4\text{SO}_2(\text{CH}_2)_{11}\text{CH}_3$	149°C
$\text{CH}_3(\text{CH}_2)_{17}\text{SO}_2(\text{CH}_2)_2\text{SO}_2(\text{CH}_2)_{17}\text{CH}_3$	150°C
$\text{CH}_3(\text{CH}_2)_{17}\text{SO}_2(\text{CH}_2)_4\text{SO}_2(\text{CH}_2)_{17}\text{CH}_3$	148°C

The compound of the above-mentioned general formula (10) may be any one of the following.

	MP
$\text{CH}_3(\text{CH}_2)_{11}\text{NHCOCONH}(\text{CH}_2)_{11}\text{CH}_3$	124°C
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCONH}(\text{CH}_2)_{17}\text{CH}_3$	121°C

The compound of the above-mentioned general formula (11) may be any one of the following.

	MP
$\text{CH}_3(\text{CH}_2)_{10}\text{CONHNHCO}(\text{CH}_2)_{10}\text{CH}_3$	151°C
$\text{CH}_3(\text{CH}_2)_{16}\text{CONHNHCO}(\text{CH}_2)_{10}\text{CH}_3$	134°C
$\text{CH}_3(\text{CH}_2)_{16}\text{CONHNHCO}(\text{CH}_2)_{16}\text{CH}_3$	147°C
$\text{CH}_3(\text{CH}_2)_{20}\text{CONHNHCO}(\text{CH}_2)_{16}\text{CH}_3$	136°C
$\text{CH}_3(\text{CH}_2)_{20}\text{CONHNHCO}(\text{CH}_2)_{20}\text{CH}_3$	143°C

The compound of the above-mentioned general formula (12) may be any one of the following.

	MP
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCO}(\text{CH}_2)_4\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	144°C
$\text{CH}_3\text{O}(\text{CH}_2)_3\text{NHCO}(\text{CH}_2)_{11}\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	140°C
$\text{CH}_3\text{CH}_2\text{O}(\text{CH}_2)_3\text{NHCO}(\text{CH}_2)_{11}\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	135°C

The compound of the above-mentioned general formula  
(13) may be ~~any one of~~ the following.

	MP
5 $\text{CH}_3(\text{CH}_2)_{16}\text{CONH}(\text{CH}_2)_6\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	149°C

The compound of the above-mentioned general formula  
(14) may be ~~any one of~~ the following.

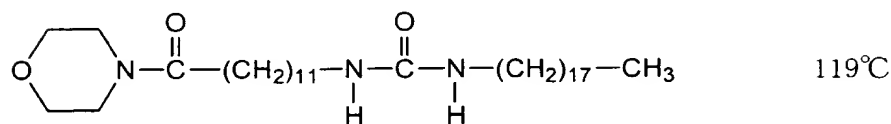
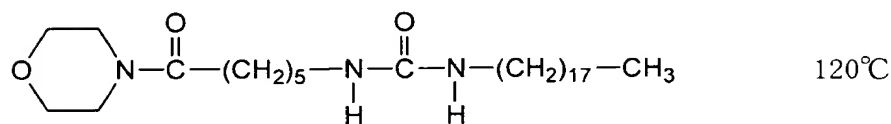
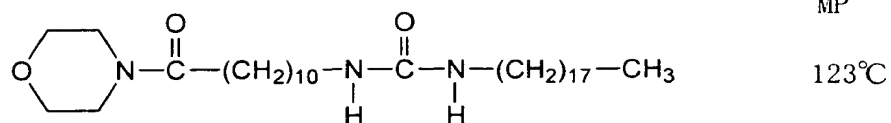
	MP
10 $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_2\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	127°C

The compound of the above-mentioned general formula  
(15) may be ~~any one of~~ the following.

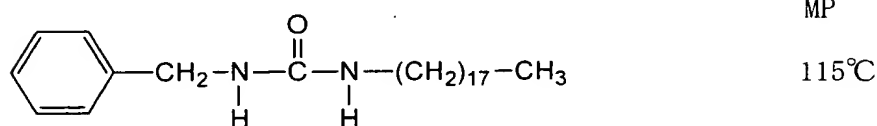
	MP
15 $\text{CH}_3(\text{CH}_2)_{17}\text{NHCONH}(\text{CH}_2)_6\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3$	177°C

The compound of the above-mentioned general formula  
(16) may be any one of the following.



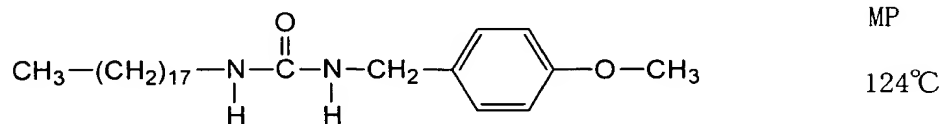


The compound of the above-mentioned general formula (17) may be ~~any one of~~ the following.



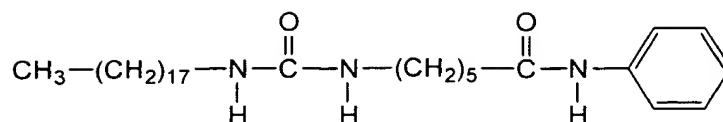
5

The compound of the above-mentioned general formula (18) may be ~~any one of~~ the following.



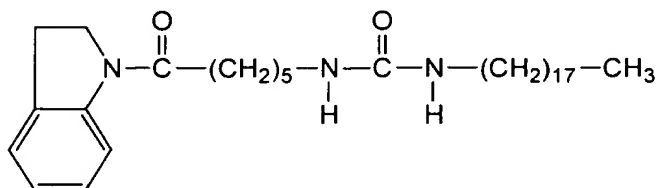
10

The compound of the above-mentioned general formula (19) may be ~~any one of~~ the following.



MP  
146°C

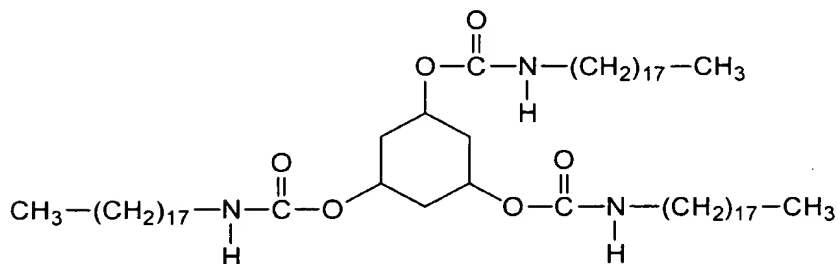
The compound of the above-mentioned general formula  
(20) may be ~~any one of~~ the following.



MP  
136°C

5

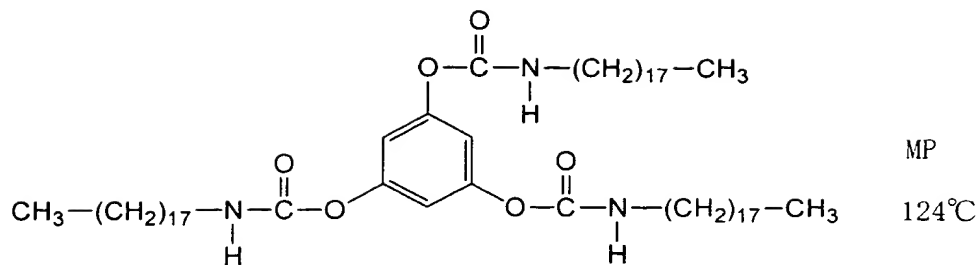
The compound of the above-mentioned general formula  
(21) may be ~~any one of~~ the following.



MP  
115°C

10

The compound of the above-mentioned general formula  
(22) may be ~~any one of~~ the following.



Examples of the method of preparing the above-mentioned linear hydrocarbon-containing compound (A)

5 include, but not limited to, the following methods.

Synthesis of compound in the general formula (7):

Synthesis of  $[\text{CH}_3(\text{CH}_2)_{17}\text{OOCNH}(\text{CH}_2)_6\text{NHCOO}(\text{CH}_2)_{17}\text{CH}_3]$

A solution of 20.1 g of stearyl alcohol and 5.1 g of hexamethylene diisocyanate in 125.5 g of tetrahydrofuran  
 10 was stirred for 3 hours under reflux. The deposited ~~crystal~~  
~~was~~ crystals were filtrated, and re-crystallized from toluene, to obtain 17.7 g of the intended compound.

Synthesis of compound in the general formula (8):

Synthesis of  $[\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_4\text{OOCNH}(\text{CH}_2)_{17}\text{CH}_3]$

15 A solution of 2.6 g of 1,4-butanediol and 18.0 g of steary isocyanate in 103.0 g of tetrahydrofuran was stirred for 5 hours under reflux. The deposited ~~crystal~~ was crystals  
~~were~~ were filtrated, and re-crystallized from toluene, to obtain 17.5 g of the intended compound.

20 Synthesis of compound in the general formula (9):

Synthesis of  $[\text{CH}_3(\text{CH}_2)_{17}\text{SO}_2(\text{CH}_2)_2\text{SO}_2(\text{CH}_2)_{17}\text{CH}_3]$

To a solution of 35.5 g of stearyl mercaptane and 8.5 g of potassium hydroxide in 177.5 g of ethanol was added 11.1 g of 1,2-dibromoethane ~~dropwise~~ drop-by-drop at room temperature, and stirred for 5 hours under reflux. After completion of stirring, 275 g of a 0.8% hydrochloric acid aqueous solution was added at room ~~temperetue~~ temperature.

The deposited ~~crystal was~~ crystals were filtrated, washed with water and dried to obtain 18.4 g of 1,2-dioctadecylthioethane.

10 Then, a mixture of 18.4 g of 1,2-dioctadecylthioethane, 184 g of acetic acid and 184 g of hydrogen peroxide (35%) was stirred at 80 to 90 °C for 5 hours. The reaction solution was added into ion exchange water under room temperature, and the deposited ~~crystal was~~ crystals were filtrated, and  
15 re-crystallized from toluene, to obtain 10.6 g of the intended compound.

Synthesis of compound in the general formula (10):

Synthesis of  $[\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOONH}(\text{CH}_2)_{17}\text{CH}_3]$

To a solution of 53.5 g of stearylamine and 15.7 g  
20 of pyridine in 599.2 g of tetrahydrofuran was added under room temperature a solution of 12.0 g of oxalyl chloride in 120.0 g of tetrahydrofuran dropwise. After stirring, the mixture was stirred at room temperature for 5 hours. The deposited ~~crystal was~~ crystals were filtrated, washed  
25 with water and re-crystallized from toluene, to obtain 19.6

g of the intended compound.

Synthesis of compound in the general formula (11):

Synthesis of  $[\text{CH}_3(\text{CH}_2)_{16}\text{CONHNHCO}(\text{CH}_2)_{17}\text{CH}_3]$

To a solution of 20.0 g of stearic hydrazide, 21.0 g of stearic acid, 10.3 g of 1-hydroxybenzotriazole in 205.0 g of tetrahydrofuran was added 9.3 g of diisopropylcarbodiimide dropwise at room temperature, and the mixture was stirred for 3 hours under reflux. The deposited ~~crystal~~ crystals were filtrated, and re-crystallized ~~from~~ using isopropyl alcohol, to obtain 23.9 g of the intended compound.

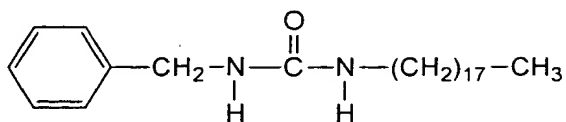
Synthesis of compound in the general formula (12):

Synthesis of  $[\text{CH}_3\text{CH}_2\text{O}(\text{CH}_2)_3\text{NHCO}(\text{CH}_2)_{11}\text{NHCONH}(\text{CH}_2)_{17}\text{CH}_3]$

A solution of 30.5 g of octadecyl isocyanate and 21.4 g of 12-aminododecanoic acid in dimethylformamide was stirred at 60 to 70 °C for 8 hours. The deposited ~~crystal~~ crystals were filtrated, and re-crystallized from toluene, and to a solution of 45.2 g of the resulted compound, 12.7 g of ethoxypropylamine and 15.6 g of 1-hydroxybenzotriazole in methylethyl ketone was added 17.8 g of diisopropylcarbodiimide dropwise at 50 to 60 °C and the mixture was stirred for 5 hours. The deposited ~~crystal~~ crystals were filtrated, and re-crystallized ~~from~~ using isopropyl alcohol to obtain 46.1 g of the intended compound.

Synthesis of compound in the general formula (17):

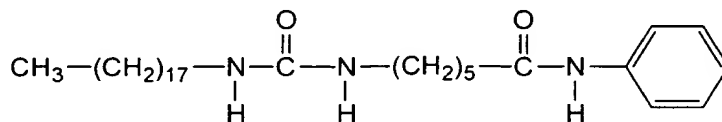
Synthesis of



5        A solution of 14.8 g of octadecyl isocyanate and 6.1 g of benzylamine in methyl ethyl ketone was stirred for 6 hours under reflux. The deposited ~~crystal was~~ crystals were filtrated, and re-crystallized ~~from~~ using isopropyl alcohol to obtain 17.2 g of the intended compound.

10      Synthesis of compound in the general formula (19):

Synthesis of

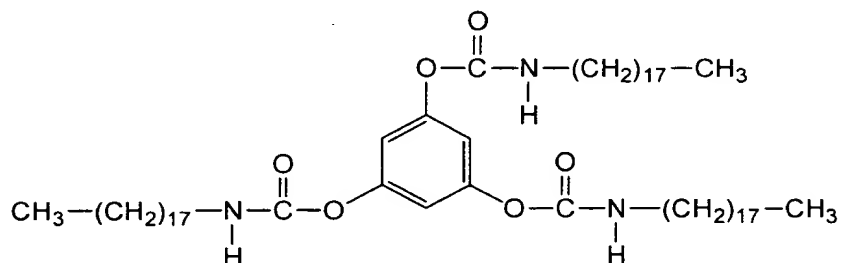


15        A solution of 14.1 g of octadecyl isocyanate and 6.2 g of 6-aminocaproic acid in dimethylformamide was stirred for 5 hours at 50 to 60 °C. The deposited ~~crystal was~~ crystals were filtrated, and re-crystallized ~~from~~ using toluene, and to a solution of 18.7 g of the resulted compound, 5.0 g of aniline ad 6.5 g of 1-hydroxybenzotriazole in methyl ethyl  
20      ketone was added 5.5 g of diisopropylcarbodiimide dropwise, and the mixture was stirred for 5 hours under reflux. The

deposited ~~crystal was~~ crystals were filtrated, and re-crystallized ~~from~~ using ethyl alcohol to obtain 16.3 g of the intended compound.

Synthesis of compound in the general formula (22):

5        Synthesis of



A solution of 4.3 g of benzene-1,3,5-triole and 33.4 g of octadecyl isocyanate in tetrahydrofuran was stirred  
10 for 8 hours under reflux. The deposited ~~crystal was~~ crystals were filtrated, and re-crystallized ~~from~~ using toluene to obtain 25.6 g of the intended compound.

Another object of the present invention is solved by further using a linear hydrocarbon-containing ~~compounds~~  
15 compound (B) having a melting point lower than the melting point of the linear hydrocarbon-containing compound (A) by 20 °C or more and having no carboxyl group, in addition to the linear hydrocarbon-containing compound (A), as the organic lower molecular weight substance.

20        This linear hydrocarbon-containing compound (B) may be one compound or a combination of two or more compounds.

The melting temperature of the linear hydrocarbon-containing compound (B) ~~is~~ should preferably be 50 °C or more and less than 100 °C. ~~This~~ The melting point ~~is~~ should more preferably be 60 °C or more, further  
5 preferably 70 °C or more, and more particularly preferably 90 °C or less.

If the melting point is too low, the image does not withstand heat resistance of an image lowers. On the other hand, ~~and~~ when the melting point is too high, the ~~clearing~~  
10 transparent-state temperature range cannot be ~~enlarged~~ widened, and ~~deleting property decreased, undesirably the~~ image cannot be erased effectively.

The mixing ratio by weight of the linear hydrocarbon-containing compound (A) to the linear  
15 hydrocarbon-containing compound (B) ~~is~~ should preferably be from 80:20 to 1:99.

~~Regarding this mixing ratio, the~~ The proportion of the linear hydrocarbon-containing compound (B) ~~is~~ should more preferably be 97 or less, further preferably 95 or less,  
20 and more particularly preferably 90 or less. Further, ~~it~~ is the proportion of the compound (B) should preferably be 30 or more, ~~more~~ further preferably 40 or more, ~~further~~ and particularly preferably 50 or more.

The linear hydrocarbon-containing compounds (A) and  
25 (B) may be used each alone or in admixture of two or more.



When the proportion of the linear hydrocarbon-containing compound (B) is too high, even in the ~~clearing~~ transparent-state temperature range, such difference in transparency occurs that transparency is higher ~~at lower~~ when the temperature side is low and transparency is lower ~~at higher~~ when the temperature side, ~~consequently,~~ is high. As a result uniform ~~clearing~~ transparency can not be accomplished, ~~undesirably.~~

~~When~~ On the other hand, when the proportion of the linear hydrocarbon-containing compound (B) is too low, sufficient transparency cannot be obtained, ~~undesirably.~~

The linear hydrocarbon-containing compound (B) preferably has no hydroxyl group, and preferably has a methyl group at the end of the molecule.

The linear hydrocarbon-containing compound (B) may be, but are not limited to, fatty esters, ketones having a higher alkyl group, dibasic acid esters, polyhydric alcohol difatty esters, aliphatic monoamide compounds, aliphatic monourea compounds.

Specific examples of the linear hydrocarbon-containing compound (B) are listed below.

The fatty ester may be octadecyl laurate, dococyl laurte, dococyl myristate, dodecyl palmitate, tetradecyl palmitate, pentadecyl palmitate, hexadecyl palmitate, octadecyl palmitate, triacontyl palmitate, octadecyl

palmitate, docodyl palmitate, vinyl stearate, propyl stearate, isopropyl stearate, butyl stearate, amyl stearte, heptyl stearate, octyl stearate, tetradecyl stearate, hexadecyl stearate, heptadecyl stearate, octadecyl  
5 stearate, dococyl stearate, hexaconyl stearate, triacontyl stearate, dodecyl behenate, octadecyl behenate, dococyl behenate, tracocyl lignocerate, myricyl melissate.

The ketone having a higher alkyl group may be 8-pentadecanone, 9-heptadecanone, 10-nonadecanone,  
10 11-heneicosanone, 12-tricosanone, 14-heptadosanone, 16-hentriacontanone, 18-pentatriacontanone, 22-tritetracontanone, 2-pentadecanone, 2-hexadecanone, 2-heptadecanone, 2-octadecanone, 2-nonadecanone.

The dibasic acid ester is preferably diesters, and  
15 it is represented by the following general formula (I).

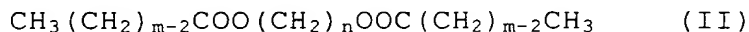


~~(wherein~~ In this formula, R and R' represent an alkyl group,  
20 and the carbon number of this alkyl group ~~is~~ should preferably be from 1 to 30, and more preferably from 1 to 22. R and R' may be the same or different. n ~~is~~ should preferably be from 1 to 30, and more preferably from 2 to 20.)~~.~~

Specifically, succinic acid diester, adipic acid  
25 diester, sebacic acid diester,

1,18-octadecamethylenedicarboxylic acid ester and the like are listed.

As the polyhydric alcohol difatty ester, those ~~of~~ represented by the following general formula (II) are  
5 listed.

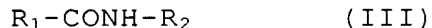


~~(wherein~~ In this formula, n is should be preferably from  
10 2 to 40, further preferably from 3 to 30, and more particularly  
preferably from 4 to 22. m is should preferably be from  
2 to 40, further preferably from 3 to 30, and more particularly  
preferably from 4 to 22.)

Specifically, 1,3-propanediol dialkanoic acid ester,  
15 1,6-hexanediol dialkanoic acid ester, 1,10-dicanediol  
dialkanoic acid ester, 1,18-octadecanediol dialkanoic acid  
ester and the like are listed.

As the fatty monoamide, those ~~of~~ represented by the  
following general formula (III) are listed.

20

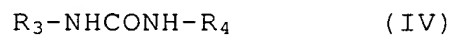


~~(wherein~~ In this formula, R<sub>1</sub> represents a linear hydrocarbon  
chain having 1 to 25 carbon atoms, R<sub>2</sub> represents a linear  
25 hydrocarbon chain having 1 to 26 carbon atoms, a methylol

group or hydrogen, and at least any one of  $R_1$  and  $R_2$  is a linear hydrocarbon chain having 10 or more carbon atoms. ~~)-~~

Specifically, N-lauryllaoric amide,  
N-palmitylpalmitic amide, N-stearylpalmitic amide,  
5 N-behenylpalmitic amide, N-palmitylstearic amide,  
N-stearylstearic amide, N-behenylstearic amide,  
N-palmitylbehenic acid, N-stearylbehenic amide,  
N-behenylbehenic amide and the like.

As the aliphatic urea compound, those ~~of~~ represented  
10 by the following general formula (IV) are listed.



~~(wherein~~ In this formula,  $R_3$  and  $R_4$  represent an alkyl group,  
15 alicyclic group or aromatic group, and at least any one of  
them is a linear hydrocarbon chain having 1 to 26 carbon  
atoms. ~~)-~~

Specifically, N-butyl-N-stearylurea,  
N-phenyl-N-stearylurea, N-stearyl-N-stearylurea,  
20 N-behenyl-N-stearylurea, N-stearyl-N-behenylurea,  
N-behenyl-N-behenylurea and the like are listed.

It is preferable that the thermo reversible recording  
medium of the present invention has the following (i) to  
(iii) together as the condition manifesting reversible  
25 change between ~~transparent condition~~ transparent-state and

~~non-transparent condition~~ opaque-state depending on temperature:

(i) the ~~clearing~~ upper limit of the transparent-state temperature is 110 °C or more,

5 (ii) the temperature difference between the ~~clearing~~ upper limit of the transparent-state temperature and the ~~opaque-state~~ lower limit of the opaque-state temperature is 20 °C or less,

(iii) the ~~clearing~~ transparent-state temperature  
10 range is 30 °C or more.

The ~~clearing~~ upper limit of the transparent-state temperature ( $T_{WX}$ ), ~~opaque-state~~ lower limit of the opaque-state temperature ( $T_{VO}$ ), temperature difference ( $\Delta T_{WV}$ ) between the ~~clearing~~ this upper limit temperature and the ~~opaque-state~~ lower limit temperature, ~~clearing-initiation~~ temperature ( $T_{WD}$ ) at which the transparent-state starts, ~~clearing and~~ the range of the transparent-state temperature range ( $\Delta T_z$ ) are determined as described below. First, an ~~opaque~~ thermo reversible recording medium is prepared.

20 ~~For of a cleared~~ If this medium is transparent or a medium which it is not sufficiently opaque, the medium is ~~opaque~~ previously required to be made opaque beforehand. The medium may be made opaque by heating, for example, by ~~pushing~~ putting the medium ~~to~~ on a hot plate  
25 ~~sufficiently heated to heat the medium.~~ The heating time

may be ~~from~~ for 10 to 30 seconds.

~~To confirm that the heating temperature is sufficiently high for the opacification,~~ It is checked visually whether the medium has become completely opaque.

5 ~~it~~ It may be advantageous to heat the medium again at somewhat higher temperature ~~somewhat higher than that temperature~~ (for example, temperature higher by 10 °C) and check the opaque-state. If degree of opacification is the same ~~in the both cases~~ before and after this heating, it  
10 ~~is known~~ confirmed that the ~~initial heating~~ temperature of previous heating is temperature sufficient to cause opacification.

If the degree of opacification is higher ~~when the medium is heated at somewhat higher temperature~~ at the previous  
15 and later heating, it ~~is known~~ will mean that the ~~initial temperature is still lower, therefore~~ medium has not become completely opaque. In this case, it may be recommendable to ~~raise heating~~ heat the medium at still higher temperature and repeat the same procedure ~~again~~ until the medium becomes  
20 completely opaque.

Then, this ~~opacified recording opaque~~ medium is heated at varied temperatures, ~~and~~ to determine the temperature at which ~~clearing occurs is checked~~ transparent-state occurs. ~~For heating of a recording medium,~~ Such heating may be  
25 performed using a heat gradient tester ~~†~~. A heat gradient

tester, for example, manufactured by Toyo Seiki K.K., HG-100~~+~~  
~~is used~~ is known. This heat gradient tester has 5 heating  
blocks, ~~and~~. The temperature in each block can be set at  
~~individual temperature differently, and~~ the heating time  
5 and pressure can also be controlled, and a medium can be  
heated at 5 different temperatures simultaneously under set  
conditions.

Specifically, the heating time is 1 second, the  
pressure in heating is about 2.5 kg/cm, and the heating  
10 temperature is changed at a constant temperature interval  
of 1 to 5 °C from lower temperature at which whiteness does  
not change by heating to temperature at which sufficient  
opacification occurs.

To prevent sticky adhesion of a medium to a heating  
15 block, a thin film (10 μm or less) of polyimide and polyamide  
may also be ~~place~~ placed on the block.

After ~~this heated~~ this heating, ~~a the~~ medium is cooled  
to room temperature, and the concentration of heated part  
at each temperature is measured using ~~Macbeth~~ Macbeth RD-914  
20 reflection densitometer, ~~and a~~. A graph is made plotted  
in which the ~~set~~ temperature ~~of a~~ set in the heat gradient  
tester is ~~recorded~~ plotted along the abscissa and the  
reflection density is ~~recorded~~ plotted along the vertical  
axis as shown in Fig. 2.

25 Fig. 2 is a graph showing the relation between the

set temperature and the reflection ~~concentration of this~~  
~~heat gradient tester density~~. ~~When~~ If the thermo reversible  
recording medium ~~uses~~ is prepared using a transparent  
substrate, then a sheet absorbing light or a sheet  
5 manifesting regular reflection of light obtained by  
vapor-depositing a metal such as Al and the like is placed  
on the rear surface of this medium.

This graph is completed, ~~by plotting concentration~~  
~~values against respective temperatures, then, connecting~~  
10 ~~the plotted respective adjacent points by straight lines~~  
joining the plotted points with a line. ~~The graph drawn~~  
As shown in Fig. 2, usually show such a graph will have a  
~~trapezoid form as shown in Fig. 2~~ trapezoidal shape.

These data ~~is~~ are influenced by the thickness and raw  
15 material of ~~a heat~~ the thermo sensible layer and the ~~medium~~  
~~including a~~ substrate.

The thickness of the medium ~~is not influenced by~~ will  
have no influence if the thickness providing it is 300  $\mu\text{m}$   
or less, and approximately the same data ~~are~~ will be obtained.  
20 On the other hand, and in the case of if the thickness ~~of~~  
is not less than 300  $\mu\text{m}$ , it may be recommendable ~~that the~~  
~~substrate side is shaved or peeled~~ to shave or peel the  
substrate to make reduce the thickness to 300  $\mu\text{m}$  or less,  
or the ~~value exceeding 300  $\mu\text{m}$  is converted~~ data may be  
25 corrected based on the thickness.



The raw material is not restricted ~~providing~~ provided it is ~~composed~~ mainly of a polymer, however. However, in the case of if the raw material is a metal and the like, conversion then appropriate correction is necessary.

5        ~~From this graph, the~~ The above-mentioned ~~clearing~~ upper limit temperature and ~~specifying~~ lower limit temperature and the like ~~are read and calculated~~ can be obtained based on the graph in Fig. 2.

First, the maximum ~~concentration value~~ reflection  
10 density ( $D_{PD}$ ) is read ~~in~~ from this graph. Then, a line of  $y = 0.7 \times D_{PD}$  is drawn, and ~~plotting points of~~ on this graph that have higher concentration density than the density represented by this line are selected.

The number of ~~the plotting~~ such points is preferably  
15 from 5 to 20.

~~When~~ If the number of ~~plotting~~ points is small, then the later calculation result becomes incorrect.

~~When~~ If the number of plotting points is small, it is necessary ~~that~~ to increase the number of points by reducing  
20 the temperature interval of at the time of heating by the above-mentioned heat gradient tester is narrow and the number is increased.

Of the selected ~~plotting~~ points, the same number of points of larger concentration values and points of smaller  
25 concentration values are removed ~~are removed~~ respectively,

and the remaining concentration values are averaged to obtain average ~~transparent concentration~~ reflection density ( $D_{WDY}$ ).

The proportion of removal of ~~larger concentration~~ high  
5 density values and ~~smaller concentration~~ low density values is from 10 to 30%, preferably from 15 to 25%, respectively, based on the selected plotting points.

By thus removing ~~large concentration values~~ and ~~smaller concentration values~~ high and low density values,  
10 the a correct value of transparent concentration reflection density of a the medium can be calculated.

Then, the ~~clearing~~ transparent-state lower limit ~~concentration~~ density ( $D_{WP}$ ) is calculated ~~according to~~ using the following numerical formula (1)

15 
$$D_{WP} = D_{WDY} - 0.2 \times (D_{WDY} - D_{PLQ}) \quad \dots (1)$$

Here,  $D_{PLQ}$  means the maximum ~~opacification~~ opaque-state concentration density, and it is calculated from the average value of ~~concentrations~~ densities of adjacent 3 points which have reached within 0.3 during  
20 raising temperature.  $D_{WP}$  represents, when at this ~~concentration density~~ or more, ~~concentration~~ a density at which the medium looks almost transparent visually. Further, a line of  $y=D_{WP}$  is drawn on the graph, and the crossing ~~pointsthereof~~ points thereof with ~~a the~~  
25 ~~concentration-temperature~~ density-temperature curve are

obtained. Of these crossing points, the one of that  
represents lower temperature side is called clearing is taken  
as the transparent-state lower limit temperature  
 (T<sub>wo</sub>) and the one of that represents higher temperature side  
 5 is called clearing is taken as transparent-state upper limit  
temperature temperature (T<sub>wx</sub>). The clearing  
transparent-state temperature range (ΔT<sub>z</sub>) is calculated  
 according to using the following numerical formula (2).

$$\Delta T_{(w)z} = T_{wx} - T_{wo} \quad \dots (2)$$

10 The ~~opacification~~ opaque-state upper limit temperature  
 (D<sub>s</sub>) is calculated according to using the following numerical  
 formula (3).

$$D_s = D_{PLQ} + 0.1 \times (D_{WDY} - D_{PLQ}) \quad \dots (3)$$

A line of y=D<sub>s</sub> is drawn on the graph, and the temperature  
 15 at a ~~crossing~~ crossing point thereof with part of a the  
~~concentration-temperature~~ density-temperature curve over  
 which transparency changes to opacification is ~~called~~  
~~opacification~~ taken as the opaque-state lower limit  
 temperature (T<sub>vo</sub>).

20 The difference (ΔT<sub>wv</sub>) between the ~~clearing~~  
transparent-state upper limit temperature and the  
~~opacification~~ opaque-state lower limit temperature is  
 calculated according using the following numerical formula  
 (4).

$$25 \quad \Delta T_{wv} = T_{vo} - T_{wx} \quad \dots (4)$$

The ~~clearing initiation concentration~~ density when the transparent-state starts ( $D_{WD}$ ) is calculated ~~according to~~ using the following numerical formula (5).

$$D_{WD} = D_{PLQ} + 0.25 \times (D_{WDY} - D_{PLQ}) \quad \dots (5)$$

5        The ~~clearing initiation~~ temperature when the transparent-state starts ( $T_{WD}$ ) is obtained from crossing points of  $y = D_{WDY} D_{WD}$  with the graph as shown in Fig. 2.

      The ~~clearing~~ transparent-state upper limit temperature ( ~~$T_{WD}$  to~~  $T_{WX}$ ) ~~is~~ should preferably be 110 °C or  
10 more, and further preferably 115 °C or more.

      When the transparent-state upper limit temperature  $T_{WX}$  rises, ~~expansion~~ widening of the ~~clearing~~ transparent-state temperature range becomes possible without lowering image durability.

15        The ~~clearing~~ transparent-state upper limit temperature ( $T_{WX}$ ) ~~is~~ should preferably be 120 °C or more, ~~more~~ further preferably 125 °C or more, ~~further~~ and more particularly preferably 130 °C or more.

      When this temperature is higher, printing sensitivity  
20 increases.

      The transparent-state upper limit temperature  $T_{WX}$  ~~is~~ should preferably be 170 °C or less, ~~more~~ further preferably 160 °C or less, ~~further~~ more particularly preferably 150 °C or less.

25        When this temperature is lower, the printing

sensitivity increases.

The difference ( $\Delta T_{wv}$ ) between the ~~clearing~~  
transparent-state upper limit temperature and the  
~~opacification~~ opaque-state lower limit temperature ~~is~~  
5 should preferably be 20 °C or less.

~~When  $T_{wv}$~~  If the difference  $\Delta T_{wv}$  is larger than this,  
temperature to cause opacification increases than necessary,  
consequently, in forming an opacified image, extremely high  
energy is necessary, and if formation of an image and deletion  
10 of the image are repeated, the surface of a medium is injured  
and the degree of opacification of the image lowers,  
~~undesirably.~~

The temperature difference  $\Delta T_{wv}$  ~~is~~ should preferably  
be 15 °C or less, more preferably 10 °C or less.

15 The ~~clearing-initiation~~ transparent-state start  
temperature ( $T_{wd}$ ) ~~is~~ should preferably be less than 95 °C,  
~~more~~ further preferably 90 °C or less, ~~further~~ and more  
particularly preferably 85 °C or less.

The transparent-state start temperature  $T_{wd}$  ~~is~~ should  
20 preferably be 70 °C or more, more preferably 75 °C or more.

When this temperature is lower, the image deleting  
~~property increases~~ can be erased effectively, and when higher,  
image durability is improved.

The ~~clearing~~ transparent-state temperature range  
25 ( $\Delta T_z$ ) ~~is~~ should preferably be 30 °C or more.

When the transparent-state temperature range  $\Delta T_z$  is smaller than this, the image deleting property decreases can not be erased effectively.

The ~~clearing~~ transparent-state temperature range  
5 ( $\Delta T_z$ ) ~~is~~ should more preferably be 40 °C or more, further preferably 45 °C or more, and more particularly preferably 50 °C or more.

When ~~this~~ the transparent-state temperature range is larger, the image deleting property increases can be erased  
10 effectively.

The transparent-state temperature range  $\Delta T_z$  ~~is~~ should preferably be 100 °C or less, ~~more~~ further preferably 90 °C or less, ~~further~~ and more particularly preferably 80 °C or less.

15 Particularly, when the ~~clearing~~ transparent-state temperature range  $\Delta T_z$  is larger, there is a merit ~~is~~ generated that ~~uniform deletion~~ the image can be erased ~~is possible~~ uniformly even if the ~~treating~~ processing speed ~~in deletion increases~~ is increased.

20 ~~In this case,~~ The transparent-state temperature range  $\Delta T_z$  ~~is~~ should preferably be 60 °C or more, and more preferably 70 °C or more. Particularly, when the ~~clearing~~ transparent-state temperature range  $\Delta T_z$  is larger, there is a merit ~~is generated~~ that ~~uniform deletion is possible~~  
25 the image can be erased uniformly even if the ~~treating~~

processing speed in deletion increases is increased.

~~In this case,~~ Thus, the transparent-state temperature range  $\Delta T_z$  is should preferably be 60 °C or more, and more preferably 70 °C or more.

5        The maximum average particle size of ~~an~~ the organic lower molecular weight substance in ~~a heat~~ the thermo sensible layer ~~is~~ should preferably be 2.5  $\mu\text{m}$  or less, and further preferably 2.0  $\mu\text{m}$  or less. This maximum average particle size represents dispersed condition of particles  
10 of an organic lower molecular weight substance dispersed in a polymer matrix resin of ~~a heat~~ the thermo sensible layer. ~~and smaller numerical value thereof indicates more~~ When this size is small it means excellent dispersed condition, meaning that is, uniform dispersion of the particles of an organic  
15 ~~lower molecular weight substance.~~ When the maximum average particle size is smaller, the initial image ~~opacified degree~~ is more opaque and the image deleting property increase can be erased more effectively, and also the contrast of the image is improved.

20        The maximum average particle size is measured by the following method. First, a recording material is embedded in an epoxy resin, and ~~a~~ an ultra thin sliced sample is made using a diamond cutter. Then, any 5 positions on the section of ~~a heat~~ the thermo sensible layer of this sample were  
25 observed using a transmission electron microscope

(hereinafter, called TEM, H-9000 manufactured by Hitachi, Ltd.) at an acceleration voltage of 300 kV, and this sample is photographed at a magnification of 10000. The number of larger particles in a 10% range based on the total number of an organic lower molecular weight substance in the area of this photograph image (10 cm x 10 cm) ~~is~~ was determined, and the average particle size is calculated from particle sizes of particles in this 10% range. This measurement is conducted on the above-mentioned 5 positions, and the average particle size is calculated on each position, ~~and the~~. The average value of 5 average particle sizes is used as the maximum average particle size. ~~In this measuring method,~~ ~~an~~ If the particle of the organic lower molecular weight substance in the area of the photograph ~~image~~ is not in uniform circular form, then the maximum particle size of particles is used. Further, when particles in the area of the photograph ~~image~~ are remarkably large and ~~the~~ their number ~~of them~~ is less than 10, the particle size of the ~~maximum particle~~ biggest one of them ~~is~~ was measured, and used as the maximum average particle size.

The substrate is not particularly restricted, and films or plates made of synthetic resins such as polyester, polyvinyl chloride, polyimide, further, metal deposited films obtained by vapor-depositing metals such as aluminum and the like on these materials, are exemplified.



The resin as ~~a~~ the mother material used in ~~a heat~~ the thermo sensible layer is a material which forms a layer in which an organic lower molecular weight substance is uniformly dispersed ~~an~~ and kept and exerts an influence on  
5 the transparency at the maximum transparency.

Therefore, ~~a~~ the resin mother material ~~is~~ preferable ~~compose of~~ includes a resin having excellent transparency, mechanical stability and excellent film forming property.

This resin preferably has a glass transition  
10 temperature preferably ~~of~~ 50 °C or more, ~~more~~ further preferably ~~of~~ 60 °C or more, ~~further~~ and more particularly preferably ~~of~~ 70 °C or more.

Further, the glass transition temperature ~~is~~ should preferably be less than 100 °C, and more preferably less  
15 than 90 °C. When the glass transition temperature is too low, the image durability decreases, ~~and when too higher.~~ When the glass transition temperature is too high, the image deleting property decreases, ~~undesirably can not be deleted effectively.~~

20 This resin has a gel proportion preferably of 30% or more, ~~more~~ further preferably of 50% or more, ~~further~~ more particularly preferably of 50% or more, and still more particularly preferably of 80% or more.

The gel proportion means a rate of production of gel  
25 when resin solutes lose independent mobility due to mutual

action in a solvent and aggregate to form solidified condition (gel).

When the gel proportion is small, repeating durability decreases, ~~therefore, for.~~ For improving the gel  
5 proportion, it may be advantageous that a hardenable resin which is hardened by heat, UV, EB and the like is ~~mized~~ mixed in a resin, or a resin itself is cross-linked.

~~For measuring~~ The gel proportion may be measured as follows,  
10 a. A film is peeled from a the substrate and the initial weight of the film is measure measured ("initial weight") ~~, then.~~ Then, the film is clamped by 400 mesh wire gauzes and immersed in a solvent in which a resin before cross-linking is soluble for 24 hours, then, dried in vacuo, and the weight after drying is measured.

15 The gel proportion is calculated ~~according to~~ using the following formula.

Gel proportion (%) = [weight after drying (g)/initial weight (g)] × 100

~~When the~~ The gel proportion is calculated ~~according to this calculation, calculation is conducted excepting by~~  
20 ~~excluding the~~ weights of the particles other than the particles of the resin in the thermo sensible layer such as the particles of an organic lower molecular weight substance, and the like other than a resin components in  
25 ~~a heat sensible layer. In this calculation, when~~ If the

weight of ~~an~~ the organic lower molecular weight substance is not previously known, it may be ~~advantageous that~~ calculated as follows. For example, area ratio occupying unit area is measured by observation of section by TEM, SEM  
5 and the like and weight ratio is calculated from specific gravities of ~~a~~ the resin and the organic lower molecular weight substance. Furthermore, ~~and~~ the weight of ~~an~~ the organic lower molecular weight substance is calculated from ~~them~~ the weight ratio, and then the ~~value of~~ gel proportion  
10 is calculated.

~~Further, in the above-mentioned measurement, when a~~  
When the reversible heat thermo sensible layer is ~~mounted~~  
formed on ~~a~~ the substrate and other layers such as ~~a~~ the  
protective layer and the like are laminated on this, or when  
15 other layer is present between ~~a~~ the substrate and ~~a heat~~  
the thermo sensible layer, ~~it may be advantageous, as~~  
~~described above, that~~ then the film thickness of ~~a~~ the  
reversible ~~heat thermo~~ sensible layer and the other layers  
~~are first checked by observation of~~ may be obtained using  
20 the method explained above, that is, by observing the section  
by TEM, SEM and the like. For example, the surface is cut  
to a depth corresponding to thickness of other layers to  
expose the surface of ~~a~~ the reversible ~~heat thermo~~ sensible  
layer, and further, the reversible ~~heat thermo~~ sensible layer  
25 is peeled, and the gel proportion is measured in the same

manner as ~~in the above-mentioned measuring method~~ explained above.

Further, when ~~a~~ the protective layer and the like made of ~~a~~ an ultraviolet ray-hardening resin and the like are placed on the top of ~~a heat~~ the thermo sensible layer, it is necessary, for preventing mixing of this layer to the utmost, to shave the surface to a depth corresponding to the thickness of ~~a the~~ the protective layer and to slightly shave also the surface of ~~a heat~~ the thermo sensible layer to prevent influence on the value of gel proportion.

It is preferable that this resin is cross-linked, ~~and~~. When the resin is ~~when~~ cross-linked, the thermo recording medium does not easily manifest change of the structure in ~~a heat~~ the thermo sensible layer even if ~~printing~~ recording and ~~deleting~~ erasing of image are repeated, and does not cause reductions in the degree of opacity, transparency and the like, meaning improvement in repeating durability.

When the resin is cross-linked, it is preferable that the resin contains a functional group such as a hydroxyl group, carboxyl group, epoxy group, acryloyl group, methacryloyl group and the like.

As the methods of cross-linking ~~method~~, there are known the methods ~~by~~ of thermal cross-linking, irradiation with UV and EB, ~~and~~ etc. Also, it is preferable to add a cross-linking agent such as an isocyanate compound,

functional acryl or methacryl monomer and the like, to cause cross-linking.

As this resin, there are exemplified vinyl chloride-based copolymers such as polyvinyl chloride, vinyl  
5 chloride-vinyl acetate copolymer, vinyl chloride-vinyl acetate-vinyl alcohol copolymer, vinyl chloride-vinyl acetate-maleic acid copolymer, vinyl chloride-acrylate copolymer and the like; vinylidene chloride-based copolymers such as polyvinylidene chloride, vinylidene  
10 chloride-vinyl chloride copolymer, vinylidene chloride-acrylonitrile copolymer and the like; polyesters; polyamides; polyacrylate or polymethacrylate or acrylate-methacrylate copolymer; silicone resins and the like.

15 These resins may be used alone or in admixture of two or more.

~~In the case~~ When the resist is a combination of a thermoplastic resin having a hydroxyl group with an isocyanate compound, it is preferable that a linear  
20 isocyanate compound and a cyclic isocyanate compound are mixed and used.

When a linear isocyanate compound is solely used, a resin cross-linked becomes usually flexible and deleting property increases, ~~however.~~ However, when ~~a heat~~ the  
25 thermo sensible layer is too soft, such a demerit as reduction

in repeating durability and image heat resistance occurs.

However, when only a cyclic isocyanate compound is used, there are ~~defect~~ defects that the cross-linked resin becomes rigid, repeating durability and image heat resistance are improved, however, deleting property lowers.

By use of a linear isocyanate compound and a cyclic isocyanate compound in admixture, deleting property, durability and heat resistance can be satisfied simultaneously.

10        The mixing ratio of ~~a~~ the linear isocyanate compound to ~~a~~ the cyclic isocyanate compound ~~is~~ should preferably be from 90:10 to 10:90, ~~more~~ further preferably from 90:10 to 30:70, ~~further~~ and more particularly preferably from 80:20 to 30:70, by weight.

15        When the proportion of a linear isocyanate compound is larger, ~~deletion~~ erasing rate and maximum ~~deletion~~ erasing tendency are improved, resulting in possibility of improvement in contrast of the image.

20        As the linear isocyanate compound, for example, those obtained by reacting a linear compound having a hydroxyl group such as triol and the like with an aliphatic isocyanate such as hexamethylene diisocyanate and the like directly or via a single or a plurality of ethylene oxides or propylene oxides are listed.

25        ~~A~~ The linear isocyanate compound has a molecular weight

preferably of 500 or more, ~~more~~ further preferably ~~of~~ 700 or more, ~~further~~ and more particularly preferably ~~of~~ 1000 or more.

Further, ~~is it~~ the molecular weight of the linear  
5 isocyanine compound should preferably be 5000 or less, ~~more~~  
further preferably 4000 or less, ~~further~~ and more  
particularly preferably 3000 or less.

When the molecular weight is too small, ~~a~~ the  
cross-linked film does not easily get a flexible structure,  
10 consequently, deleting property decreases, ~~and when~~. When  
the molecular weight is too large, molecules cannot move  
easily, consequently, the degree of cross-linking decreases  
to lower the durability.

The molecular weight per one isocyanate group ~~is~~ should  
15 preferably be 250 or more, ~~more~~ further preferably 300 or  
more, ~~further~~ and more particularly preferably 400 or more.

Further, ~~it is~~ the molecular weight per one isocyanine  
group should preferably be 2000 or less, ~~more~~ further  
preferably 1500 or less, ~~further~~ and more particularly  
20 preferably 1000 or less.

When the molecular weight per one isocyanate group  
is too small, ~~a~~ the cross-linked film does not easily get  
~~a~~ the flexible structure, consequently, deleting property  
decreases, ~~and when~~. When the molecular weight is too large,  
25 molecules cannot move easily, consequently, the degree of

cross-linking decreases to lower durability.

The cyclic ~~isocyanate compound~~ isocyanate compound is an isocyanate compound having a benzene ring or isocyanurate ring.

5        Among them, a type having an isocyanurate ring is suitably used since it shows no yellowing.

It is preferable that also ~~a~~ the cyclic isocyanate compound have a linear structure such as an alkylene chain and the like other than a cyclic structure.

10        The molecular weight of ~~a~~ the cyclic isocyanate compound ~~is~~ should preferably be 100 or more, ~~more~~ further preferably 200 or more, ~~further~~ and more particularly preferably 300 or more. Further, it ~~is~~ should preferably be less than 1000, ~~more~~ and further preferably less than  
15        700.

When this molecular weight is too small, the compound evaporates by heat in forming a film, and cross-linking of the film becomes impossible and durability decreases, ~~and when~~. When this molecular weight is too large, a rigid  
20        structure cannot be formed and durability decreases. As the mixture of a linear isocyanate compound and a cyclic isocyanate compound, the above-mentioned materials may be mixed and used, or a commercially available mixture may be used.

25        Examples of the commercially available mixture include,



but not limited to, "Coronate 2298-90T" manufactured by Nippon Polyurethane K.K., and the like.

The ~~heat~~ thermo sensible layer of the thermo reversible recording medium of the present invention has a thickness of preferably from 1 to 30  $\mu\text{m}$ , ~~more~~ further preferably from 2 to 20  $\mu\text{m}$ , ~~further~~ and more particularly preferably from 4 to 15  $\mu\text{m}$ . When the ~~recording-heat~~ thermo sensible layer is too thick, distribution of heat is generated in the layer and uniform clearing becomes difficult. When the ~~heat-~~ thermo sensible layer is too thin, the degree of opacification lowers, to decrease contrast.

The degree of opacification can be increased by increasing the amount of fatty acids in the recording layer and cross-linking the resin in the ~~heat-~~ thermo sensible layer.

The ratio by weight of an organic lower molecular weight substance to ~~a~~ the resin in ~~a~~ the ~~heat-~~ thermo sensible layer ~~is~~ should preferably be from about 2:1 to 1:16, ~~more~~ further preferably from 1:2 to 1:8, more particularly preferably from 1:2 to 1:5, still more particularly preferably from 1:2 to 1:4. ~~Most~~ and most preferable ~~is~~ from 1:2.5 to 1:4.

When the proportion of a resin is not more than the above-mentioned ~~value~~ ranges, it is difficult to form a film in which an organic lower molecular weight substance is held in ~~a~~ the resin, ~~and when~~. When the proportion is not less

than the above-mentioned ~~value~~ ranges, opacification becomes difficult since the amount of an organic lower molecular weight substance is small.

In the present invention, a protective layer can be  
5 provided on a ~~the heat-~~ thermo sensible layer for protecting the ~~heat-~~ thermo sensible layer.

The thickness of this protective layer is appropriately from 0.1 to 5  $\mu\text{m}$ , ~~and as~~. This protective layer may be prepared using the raw material thereof, there  
10 are exemplified such as silicone-based rubber or silicone resins as described in JP-A No. 63-221087, polysiloxane graft polymers as described in JP-A No. 63-317385, ultraviolet ray-curing resins or electron beam-curing resins as described in JP-A No. 02-566, and the like.

15 Further, the protective layer ~~can~~ may contain an organic or inorganic filler.

Further, an intermediate layer ~~can~~ may be provided between a ~~the~~ protective layer and a ~~heat-~~ thermo sensible layer as described in JP-A No. 1-133781, for protecting the  
20 ~~heat-~~ thermo sensible layer from a solvent and monomer components and the like in a protective layer forming solution.

~~As the raw material of the~~ This intermediate layer, may be prepared using the raw material such as resins used  
25 in a ~~heat-~~ thermo sensible layer ~~can be exemplified~~, and

in addition, thermosetting resins, thermoplastic resins, UV-curing resins and EB-curing resins ~~can~~ may be used. As such resins, polyethylene, polypropylene, polystyrene, polyvinyl alcohol, polyvinyl butyral, polyurethane, saturated polyester, unsaturated polyester, epoxy resin, phenol resin, polycarbonate, polyamide and the like are listed. The thickness of the intermediate layer ~~is~~ should preferably be from about 0.1 to 2  $\mu\text{m}$ , and when it is 0.1  $\mu\text{m}$  or more, the effect of a protective layer can be maintained, and when less than 2  $\mu\text{m}$  ~~or more~~, heat sensitivity does not lower.

Further, a ~~lay~~ showing layer achieving regular reflection of light can be provided between ~~a~~ the substrate and ~~a~~ the ~~heat~~ thermo sensible layer and the like to ~~improved~~ improve the contrast. This light reflection layer can be usually formed by a method such as vapor-deposition of metal, and the like, ~~and thickness thereof is~~. Thickness of this layer should preferably be from about 100 to 1000 ~~Armstrong~~ Ångstroms.

The thermo reversible recording label of the present invention has an adhesive layer or a sticker layer provided on the ~~opposites~~ opposite surface to a surface on which a ~~heat~~ thermo sensible layer is ~~formed of a~~ provided on the substrate ~~constituting of the above-mentioned~~ thermo reversible recording medium. The thermo reversible

recording ~~labels~~ label ~~are classified into those having~~ may  
have an adhesive layer of sticker layer formed (no-releasing  
paper type) ~~and those having~~ or may have a releasing paper  
attached under the adhesive layer of sticker layer (releasing  
5 paper type) ~~, and as the material constituting the adhesive~~  
~~layer, those of.~~ A hot melt type are material is usually  
used as the material for forming the adhesive layer.

~~As the materials of the~~ The adhesive layer or the  
sticker layer, ~~those generally used can be used~~ may be formed  
10 using known materials. Examples ~~thereof~~ of such materials  
include, but not limited to, urea resins, melamine resins,  
phenol resins, epoxy resin, vinyl acetate-based resins,  
vinyl acetate-acryl-based copolymers, ethylene-vinyl  
acetate copolymers, acrylic resins, polyvinyl ether-based  
15 resins, vinyl chloride-vinyl acetate-based copolymers,  
polystyrene-based resins, polyester-based resins,  
polyurethane-based resins, polyamide-based resins,  
chlorinated polyolefin-based resins, polyvinyl  
butyral-based resins, acrylate-based copolymers,  
20 methacrylate-based copolymers, natural cubber,  
cyanoacrylate-based resins, silicon-based resins and the  
like.

Then, a member having ~~an information memorizing part~~  
a memory and a reversible display part in which at least  
25 the above-mentioned ~~heat~~ thermo sensible layer

constituting a the thermo reversible recording medium is used as this reversible display part will be described.

~~As specific~~ The examples of ~~this~~ such a member having ~~an information memorizing part~~ the memory and ~~a the~~ reversible display part, ~~following examples are listed~~ are as follows.

(1) Members in which ~~part~~ a portion of a member having ~~an information memorizing part~~ that holds the memory is used as a substrate ~~of a thermo reversible recording medium,~~ and ~~a~~ for directly forming the heat- thermo sensible layer is directly formed of the thermo reversible recording medium.

(2) Members in which the ~~surface of a~~ substrate of ~~a the~~ thermo reversible recording medium ~~having a heat- sensible layer on the substrate,~~ is adhered to a substrate separately formed, is adhered with respect to a member having ~~an information memorizing member~~ that holds the memory.

(3) Members in which the ~~above-mentioned~~ thermo reversible recording label is adhered, via an adhesive layer or sticker layer, to a member ~~having an information~~ memorizing member that holds the memory.

(4) Members in which ~~an information memorizing part~~ the memory is provided in a the thermo reversible recording medium (~~as in this case,~~ the substrate ~~constituting a of~~ the thermo reversible recording medium, ~~those having large thickness are~~ should preferable preferably be thick).

In any case, it is necessary that ~~an information~~  
~~memorizing part~~ the memory and the reversible display part  
are so set that functions thereof can be manifested,  
respectively, ~~and in this case, the information memorizing~~  
5 ~~part can.~~ The memory may be provided on the opposite surface  
~~to a side of the surface of the substrate of the thermo~~  
reversible recording medium on which a heat the thermo  
sensible layer is provided of a substrate in a thermo  
reversible recording medium, or the memory may be provided  
10 between a the substrate and a the heat the thermo sensible layer,  
or on part a portion of a heat the thermo sensible layer.

The member ~~having this information memorizing part~~  
that holds the memory is not particular restricted, ~~and.~~  
Examples of such a member are cards, disks, disk cartridges  
15 and tape cassettes are generally exemplified.

The following examples are listed.

Thick cards such as IC cards (contact type and  
non-contact type), optical cards and the like, disk  
cartridges containing a disk which can re-write recorded  
20 information such as a floppy disk, optical magnetic recording  
disk (MD), DVD-RAM and the like, re-writable disks using  
no disk cartridge such as CR-RW, DVD-RW and the like, write  
one type disks such as CD-R and the like, video tape cassettes,  
and the like.

25 This member having both ~~of a~~ the reversible display

part and ~~an information memorizing part~~ the memory is illustrated below using a card having such a structure. By displaying part of information recorded in the information memorizing part on a ~~heat~~ thermo sensible layer, the owner  
5 of the card can recognize the information only by looking at the card without a specific apparatus, namely, the availability thereof increases extremely as compared with cards ~~using~~ having no thermo reversible recording medium.

The ~~information memorizing part~~ memory is not  
10 particularly restricted ~~only providing~~ provided it can record necessary information, ~~and for example,~~ Examples of such a memory are magnetic recording layers, contact type IC, non-contact type IC, ~~or and~~ optical memories are useful.

The magnetic recording layer is formed by applying  
15 metal compounds such as iron oxide, barium ferrite and the like usually used or resins such as vinyl-chloride-based resins, urethane-based resins and nylon-based resins on a substrate, or according to methods such as vapor deposition, sputtering and the like using the above-mentioned metal  
20 compound without using a resin.

Further, in thermo reversible recording ~~media~~ media used for display, ~~a heat~~ the thermo sensible layer can also be used as a recording part utilizing bar ~~code~~ code, two dimensional ~~code~~ code and the like.

25 ~~Regarding the~~ An example of a material using the

above-mentioned thermo reversible recording label (3) ~~in~~  
is the case of a thick substrate on which application of  
~~a heat~~ the thermo sensible layer is difficult. ~~An~~ In this  
case, the adhesive layer or sticker layer can be provided  
5 on the whole ~~surface of part of it or a portion of this material.~~  
By this, ~~availabilities of this medium can be improved, such~~  
~~as display of part~~ becomes convenient because a portion of  
information recorded on a magnetic medium, ~~and the like can~~  
be displayed.

10 This thermo reversible recording label having an  
adhesive layer or sticker layer provided can be applied not  
only to the above-mentioned magnetic vinyl chloride cards,  
but also to thick cards such as IC cards, optical cards and  
the like.

15 Further, this thermo reversible recording label can  
be used instead of a display label on a disk cartridge  
containing a disk which can re-write recorded information,  
such as a floppy disk, MD, DVD-RAM and the like.

Fig. 3 shows an example in which the thermo reversible  
20 recording label is pasted on MD disk cartridge.

Further, in ~~the~~ case of a disk using no ~~disk~~ cartridge,  
such as CR-RW and the like, it is also possible to paste  
~~a~~ the thermo reversible recording label directly on ~~a~~ the  
disk, and to provide ~~a~~ the ~~heat~~ thermo sensible layer  
25 directly on ~~a~~ the disk.



Thus, applications to uses such as automatic alteration of display contents corresponding to alteration of recorded contents, and the like, are possible.

Fig. 4 shows an example in which the thermo reversible recording label is directly pasted on CD-RW.

In the thermo reversible recording label of the present invention, it is also possible that a thermo reversible recording medium is pasted on a once write type disk such as CD-R and the like and part of recorded information once-written in CD-R can also be re-written and displayed.

Fig. 5 is an example having a constitution in which ~~a~~ the thermo reversible recording label is formed on an optical information memorizing medium (CD-RW) using AgInSbTe-based phase-variation type recording material. In a basic constitution, a first dielectric layer, ~~optical~~ optical information memorizing layer, second dielectric layer, reflection heat radiation layer and intermediate layer on a substrate having guide grooves, and a hard coat layer is provided on the rear surface of the substrate. Further, ~~a~~ the thermo reversible recording label is pasted on the intermediate layer.

Though a dielectric layer is not necessarily required to be provided on both ~~side of a~~ sides of the recording layer, when ~~a~~ the substrate is made of a material having low heat resistance such as a polycarbonate resin, it is desirable

to provide a the first dielectric layer.

Further, as shown in Fig. 6, it can also be used as a display label for a video tape cassette.

As the method of providing a thermo reversible  
5 recording function on a thick card, disk cartridge and disk,  
there are a method of directly applying a ~~heat-~~ thermo  
sensible layer on them, a method in which a ~~heat-~~ thermo  
sensible layer is previously formed on a separate substrate  
and the ~~heat-~~ thermo sensible layer is transferred onto a  
10 thick card, disk cartridge and disk, and other methods, in  
addition to the above-mentioned method of pasting a thermo  
reversible recording label.

~~In the case of transfer~~ When transferring the thermo  
sensible layer onto the thick card, an adhesive layer and  
15 sticker layer of hot melt type and the like may also be provided  
on a ~~heat-~~ the thermo sensible layer.

When a the thermo reversible recording label is pasted  
or a ~~heat-~~ the thermo sensible layer is provided on a rigid  
substrate such as a thick card, disk, disk cartridge, tape  
20 cassette and the like, it is preferable to provide a layer  
or sheet having elasticity and acting as cushion between  
the rigid substrate and a label or a ~~heat-~~ the thermo sensible  
layer, to obtain improved contact property with a thermal  
head, and to form an image uniformly.

25 ~~For example, the~~ The reversible ~~heat-~~ thermo sensible

recording medium of the present invention can be processed into a film ~~in which~~ in the manner as shown in, for example, Fig. 7A to Fig. 7C. In Fig. 7A ~~a the~~ reversible heat-thermo sensible recording layer (13) and ~~a the~~ protective layer (14) are provided on ~~a the~~ substrate (11) as shown in Fig. 7A. In Fig. 7B, ~~a film in which an the~~ aluminum reflection layer (12), a the reversible heat-thermo sensible recording layer (13) and ~~a the~~ protective layer (14) are provided on ~~a the~~ substrate (11) as shown in Fig. 7B. As shown in Fig. 7C, ~~a film in which an the~~ aluminum reflection layer (12), a the reversible heat-thermo sensible recording layer (13) and ~~a the~~ protective layer (14) are provided on ~~a the~~ substrate (11) and ~~a the~~ magnetic recording layer (16) is provided on the rear surface of the substrate (11) as shown in Fig. 7C, ~~and a.~~ Such a film may be provided on the card (21) having a printed display part (23) as shown in Fig. 8.

Further, as shown in Fig 9A for example, it is possible that a film having ~~an the~~ aluminum reflection layer (12), a the reversible heat-thermo sensible recording layer (13) and ~~a the~~ protective layer (14) provided on ~~a the~~ substrate (11) (see Fig. 7B) is processed into a card, a concave part (23) is formed to accommodate an IC tip, and processing into a card is effected.

In this example, a re-writing recording part (24) is label-processed on ~~a the~~ reversible heat-thermo sensible

recording medium in the form of ~~a~~ the card, and ~~a~~ the concave part for burying the IC tip (23) is formed on a given position on the rear surface of the reversible ~~heat~~ thermo sensible recording medium, and in this concave part (23), ~~a~~ the wafer  
5 (231) as shown in Fig. 9B is integrated and fixed.

In the wafer (231), ~~an~~ the integrated circuit (233) is provided on ~~a~~ the wafer substrate (232) and a plurality of contact terminals (234) connected electrically to this integrated circuit (233) ~~is~~ are provided on the wafer  
10 substrate (232).

This contact terminal (234) is exposed to the rear side of the wafer substrate (232), and a dedicated printer (readerwriter) is in electrical contact with this contact terminal (234), to form a constitution so that reading and  
15 re-writing of given information are possible.

Examples of use of the reversible ~~heat~~ thermo sensible recording card are illustrated in Fig. 10A and Fig. 10B.

Fig. 10A is a schematic constitution block view showing the integrated circuit (233), and Fig. 10B is a constitution  
20 block view showing one example of recording data of RAM.

The integrated circuit (233) is constituted, for example, of LSI, and it contains the CPU (235) which can execute control motion in a given order, ROM (236) accommodating motion program data of the CPU (235), and the  
25 RAM (237) which can write and read necessary data.

Further, the integrated circuit (233) contains an input ~~out-put~~ output interface (238) which receives an input signal and ~~impart~~ imparts the input data to the CPU (235) and receives an output signal from the CPU (235) and outputs  
5 the signal, and though not shown, a power on reset circuit, a clock generating circuit, pulse dividing circuit (interrupt pulse generation circuit), and address decoder circuit.

The CPU (235) can execute motion of an interrupt control  
10 routine, corresponding to an interrupt pulse imparted periodically from the pulse division circuit.

Further, the address decoder circuit decodes address data from the CPU (235), and imparts signals to the ROM (236), RAM (237) and input output interface (238), respectively.

~~To the input output interface (238) is connected a~~  
15 A plurality (8 in the figure) of contact terminals (234) are connected to the input output interface, and given data.  
Data from the above-mentioned dedicated printer (readerwriter) is input from ~~this~~ these contact ~~terminal~~  
20 terminals (234) via the input output interface (238) into the CPU (235). The CPU (235) responds to ~~an~~ the input signal and conducts each ~~motion~~ function according to program data accommodated in the ROM (236), and outputs given data and signals to the card readerwriter via the input output  
25 interface (238).

As shown in Fig. 10B, the RAM (237) contains a plurality of memory regions (239a) to (239f).

For example, card numbers are memorized in the region (239a), and ~~the~~ personal data such as the name, address, telephone number and the like of the card owner is, for example, memorized in the region (239b), and information corresponding to remaining valuable value or valuable paper which can be used by the owner is, for example, memorized in the region (239c), and information corresponding to remaining valuable value or valuable paper used by the owner is memorized in the regions (239d), (239e), (239f) and (239g).

The present invention further provides an image processing method of conducting formation and/or deletion of an image by heating using the above-mentioned thermo reversible recording medium, the above-mentioned member having ~~an information memorizing part~~ the memory, or the above-mentioned label, ~~and~~. This invention also provides an image processing apparatus for conducting formation and/or deletion of an image by heating, having the above-mentioned thermo reversible recording medium, the above-mentioned member having an information memorizing part, or the above-mentioned label.

For formation of images, ~~an image recording means are~~ unit is used which can heat a portion of the medium ~~partially~~

~~on images such as~~ in the form of an image. This image recording unit may be a thermal head, laser and the like.

For deletion of image, a hot stamp, ceramic heater, heat roller, hot air and the like and an image deletion means  
5 unit such as ~~a the~~ thermal head, laser and the like are used.

Among them, ~~a the~~ ceramic heater is preferably used.  
~~By use of a~~ When the ceramic heater is used, the overall  
size of ~~an the~~ apparatus can be ~~decreases~~ made smaller, and  
stable deletion condition can be obtained, leading to an  
10 image having excellent contrast.

The ~~setting~~ temperature ~~of a set in the~~ ceramic heated  
heater is should preferably be 100 °C or more, ~~more~~ further  
preferably 110 °C or more, ~~further~~ and particularly more  
preferably 115 °C or more.

15 Further, ~~by use of a~~ when the thermal heat head is  
used as the image deletion ~~means~~ unit, the overall size of  
the ~~whole~~ apparatus can be further ~~decreased~~ made smaller.

Furthermore, consumption of electric power ~~power~~ can  
be reduced, and a handy type apparatus of battery drive can  
20 also be used.

If one thermal head is commonly used for formation  
and deletion ~~simultaneously of image,~~ further decrease in  
the overall size is possible can be still further reduced.

When formation and deletion are conducted by one  
25 thermal head, it may be permissible that all of the previous

images are deleted once, and new images are formed renewedly,  
and an ~~over-write~~ over-write mode is also possible in which  
previous images are once deleted and new images are formed  
while changing energy for each image.

5        In the ~~over-write~~ over-write mode, the total time  
including formation and deletion is shortened, leading to  
speed up of recording.

~~When a~~ If the card having a has the heat-thermo sensible  
layer and an ~~information memorizing part is used~~ the memory,  
10    then the ~~above-mentioned~~ apparatus also ~~contains~~ includes  
means for reading and ~~means for~~ re-writing information in  
the ~~information memorizing part~~ memory.

Fig. 11 shows an example of the image processing  
apparatus of the present invention. In this apparatus,  
15    image deletion is conducted by a ceramic heat and image  
formation is conducted by a thermal head.

First, information memorized in a magnetic recording  
layer of a recording medium is read by a magnetic head, then,  
images recording in a reversible ~~heat-thermo~~ thermo sensible layer  
20    are deleted by heating by a ceramic heater, further, a new  
information treated based on the information read by the  
magnetic head is recorded by a thermal head in a reverse  
~~heat-thermo~~ thermo sensible layer. Then, the information in the  
magnetic recording layer is also re-written by new  
25    information.



Namely, in the image processing apparatus of Fig. 11,  
a the thermo reversible recording medium (1) having a  
magnetic recording layer provided on the opposite side of  
a ~~heat~~ thermo sensible layer is transported along a  
5 transporting path shown by arrows, or transported to the  
reverse direction in an apparatus along the transporting  
path.

In the thermo reversible recording medium (1),  
magnetic recording or record deletion is effected on a  
10 magnetic recording layer between a the magnetic head (34)  
and a the transportation roller (31), and heated for deleting  
images between a ceramic heater (38) and a transportation  
roller (40), and images are formed between the thermal head  
(53) and the transportation roller (47), then, transported  
15 out of the apparatus.

Re-writing of magnetic recording may be performed  
before or after deletion of images by a the ceramic heater.

Further, if desired, after passing between the ceramic  
heater (38) and the transportation roller (40), or after  
20 passing between the thermal head (53) and the transportation  
roller (47), the medium is transported to reverse direction  
along the transportation path, and thermal treatment can  
be effected again by the ceramic heater (38) and printing  
treatment can be performed again by the thermal head (53).

25 Example 1

Al layer of thickness about 400 ~~Armstrong~~ Ångstroms was vacuum vapor-deposited on the PET film side of an original drapery manufactured by Dainippon Ink & Chemicals, Ink. (Memory Dick, DS-1711-1040: a magnetic recording layer and  
5 a self cleaning layer have been applied on a transparent PET film having a thickness of 188  $\mu\text{m}$ ), to form an optical reflection layer.

On this was applied a solution composed of  
Vinyl chloride-vinyl acetate-phosphate copolymer  
10 (manufactured by Denki Kagaku Kogyo K.K., Denka Vinyl #1000P)

10 parts

Methyl ethyl ketone 45 parts

Toluene 45 parts

, and the solution was dried by heating to provide an adhesion  
15 layer having a thickness of about 0.5  $\mu\text{m}$ .

Then, into a resin dissolved solution prepared by dissolving 26 parts of a vinyl chloride-based copolymer (manufactured by Nippon Xeon Co., Ltd., MR110) into 230 parts of methyl ethyl ketone was added

20 Behenyl behenate (test sample manufactured by Miyshi Yushi Kabushiki Kaisha) 6 parts

$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCNH}(\text{CH}_2)_{17}\text{CH}_3$  (test sample manufactured by Miyshi Yushi Kabushiki Kaisha) 4 parts

, ceramic beads having a diameter of 2mm were placed in a  
25 glass bottle, and using a paint shaker (manufactured by Asada

Tekko K.K.), and the mixture was dispersed for 48 hours to prepare a uniform dispersion.

Into this dispersion, 4 parts of an isocyanate compound (manufactured by Nippon Polyurethane K.K., Coronate 2298-90T) ~~was~~ were added to prepare a ~~heat-~~ thermo sensible layer solution, and applied on ~~an~~ the adhesive layer of a PET film having the above-mentioned magnetic recording layer, and dried by heating, then, the applied solution was further kept at 60 °C for 72 hours to cross-link the resin to provide  
10 a ~~heat-~~ thermo sensible layer having a thickness of about 10 µm.

On this ~~heat-~~ thermo sensible layer was applied a solution composed of

75% solution of urethane acrylate-based ultraviolet  
15 ray-curing resin in butyl acetate (manufactured by Dainippon Ink & Chemicals, Ink, Unidick C7-157) 10 parts

Isopropyl alcohol 10 parts

, the solution was dried by heating, then, irradiated with ultraviolet ray using a high pressure mercury lamp of 80  
20 w/cm to cure the solution to form a protective layer having a thickness of about 3 µm, giving a ~~a~~ the thermo reversible recording medium.

#### Example 2

A thermo reversible recording medium was produced in  
25 the same manner as in Example 1 except that the amount of

behenyl behenate was changed to 8 parts and the amount of  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was changed to 2 parts.

#### Example 3

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 9 parts and the amount of  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was changed to 1 part.

#### Example 4

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 9.5 parts and the amount of  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was changed to 0.5 parts.

#### Example 5

A thermo reversible recording medium was produced in the same manner as in Example 2 except that behenyl behenate was replaced by diheptadecylketone (manufactured by Nippon Kasei Chemical Co., Ltd., Wax KS).

#### Example 6

A thermo reversible recording medium was produced in the same manner as in Example 2 except that behenyl behenate was replaced by ethanolamine distearate (manufactured by Nippon Kasei Chemical Co., Ltd., Suliaid S).

#### Example 7

A thermo reversible recording medium was produced in the same manner as in Example 4 except that

$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCNH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by  $\text{CH}_3(\text{CH}_2)_{16}\text{CONHNHCO}(\text{CH}_2)_{16}\text{CH}_3$  (test sample manufactured by Miyshi Yushi Kabushiki Kaisha).

Example 8

5 A thermo reversible recording medium was produced in the same manner as in Example 7 except that behenyl behenate was replaced by diheptadecylketone (manufactured by Nippon Kasei Chemical Co., Ltd., Wax KS).

Example 9

10 A thermo reversible recording medium was produced in the same manner as in Example 7 except that behenyl behenate was replaced by ethanolamine distearate (manufactured by Nippon Kasei Chemical Co., Ltd., Suliaid S).

Example 10

15 A thermo reversible recording medium was produced in the same manner as in Example 4 except that  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCNH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by  $\text{CH}_3(\text{CH}_2)_{17}\text{OOCNH}(\text{CH}_2)_6\text{NHCOO}(\text{CH}_2)_{17}\text{CH}_3$ . (test sample manufactured by Miyshi Yushi Kabushiki Kaisha).

20 Example 11

A thermo reversible recording medium was produced in the same manner as in Example 2 except that  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCNH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOO}(\text{CH}_2)_4\text{OOCNH}(\text{CH}_2)_{17}\text{CH}_3$  (test sample manufactured by Miyshi Yushi Kabushiki Kaisha).

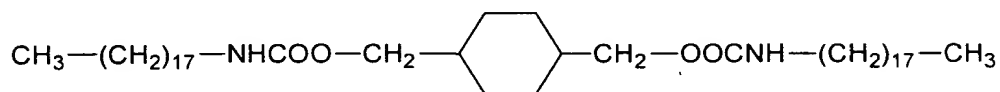
25

Example 12

A thermo reversible recording medium was produced in the same manner as in Example 4 except that  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by  $\text{CH}_3(\text{CH}_2)_{17}\text{SO}_2(\text{CH}_2)_2\text{SO}_2(\text{CH}_2)_{17}\text{CH}_3$  (test sample manufactured by Miyshi Yushi Kabushiki Kaisha).

Example 13

A thermo reversible recording medium was produced in the same manner as in Example 2 except that  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by the following material (test sample manufactured by Miyshi Yushi Kabushiki Kaisha).



Example 14

An adhesive layer, ~~heat-~~ thermo sensible layer and protective layer were formed on the Al vapor-deposited surface of an Al deposited polyester film (manufactured by Toray Industries, Inc., #50 Mtalmy) having a thickness of about 50  $\mu\text{m}$  in the same manner as in Example 1.

Further, an acrylic sticker layer having a thickness of about 5  $\mu\text{m}$  was provided on the rear surface of the ~~heat-~~ thermo sensible layer of the substrate, to produce a thermo reversible recording label.

This label was ~~made~~ cut into a donut shape as shown in Fig. 4, and pasted on CD-RW as shown in Fig. 5, to produce an optical information memorizing medium equipped with a reversible display function.

5        Using the optical information memorizing medium as produced above, part of information (date, time and the like) memorized by a CD-RW drive (manufactured by Ricoh Co., Ltd., MP6200S) was displayed and recorded on the ~~heat~~ thermo sensible layer using a recording apparatus having a recording  
10 ~~means~~ unit (thermal head) and a deleting ~~means~~ unit (ceramic heater) while controlling recording energy of the thermal head corresponding to change in recording temperature of each medium.

Further, information of the memorizing layer of the  
15 optical information memorizing medium was re-written using this drive, the previous record was deleted using the deleting ~~means~~ unit, and the re-written information was re-written newly ~~by a thermal head~~ on the ~~heat~~ thermo sensible layer using the thermal head, to effect display  
20 recording.

Further, re-writing of this display recording was repeated 100 times, however, recording and deletion were possible.

#### Example 15

25        The thermo reversible recording label in Example 14

was pasted on MD disk cartridge as shown in Fig. 3.

Part of information (date, song title and the like) memorized by MD was displayed and recorded on the ~~heat-~~ thermo sensible layer using a recording apparatus having ~~a~~ the recording ~~means~~ unit (thermal head) and ~~a~~ the ~~deleting means~~ erasing unit (ceramic heater) while controlling recording energy of the thermal head corresponding to change in recording temperature of each medium.

Further, re-writing of this display recording was repeated 100 times, however, recording and deletion were possible.

#### Comparative Example 1

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the application solution of the ~~heat-~~ thermo sensible layer was changed as follows.

Behenic acid (reagent manufactured by SIGMA, purity 99%)	5 parts
Eicosandioic acid (manufactured by Okamura Seiyu K.K., SL-20-90)	5 parts
Vinyl chloride-vinyl acetate copolymer (manufactured by Union Carbide, VYHH)	38 parts
Tetrahydrofuran	210 parts
Toluene	20 parts



## Comparative Example 2

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the application solution of the ~~heat~~-thermo sensible layer was changed as follows.

The ~~heat~~-thermo sensible layer herein formed had poor uniformity showing remarkable white particles.

Behenyl behenate (reagent manufactured by SIGMA)

9.5 parts

10 Ethylenebisbehenic amide (manufactured by Nippon Kasei Chemical Co., Ltd., Sulipax B)

0.5 parts

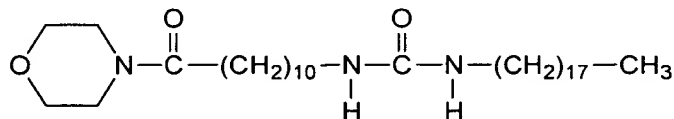
Vinyl chloride-vinyl acetate copolymer (manufactured by Union Carbide, VYHH)

15 30 parts

Tetrahydrofuran 160 parts

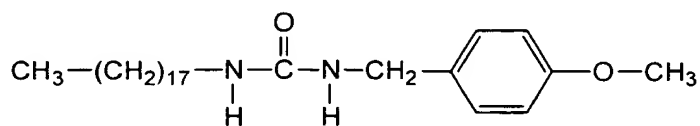
## Example 16

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCO}(\text{NH}(\text{CH}_2)_{17}\text{CH}_3)$  was replaced by 3 parts by weight of the following compound.



### Example 17

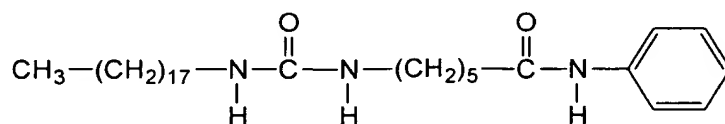
A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by 3 parts by weight of the following compound.



10

### Example 18

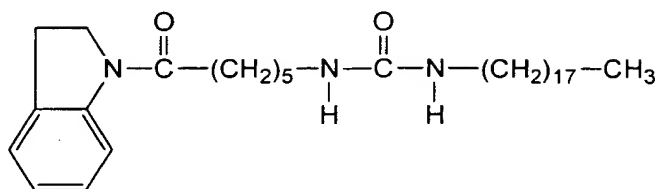
A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by 3 parts by weight of the following compound.



### 20 Example 19

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and

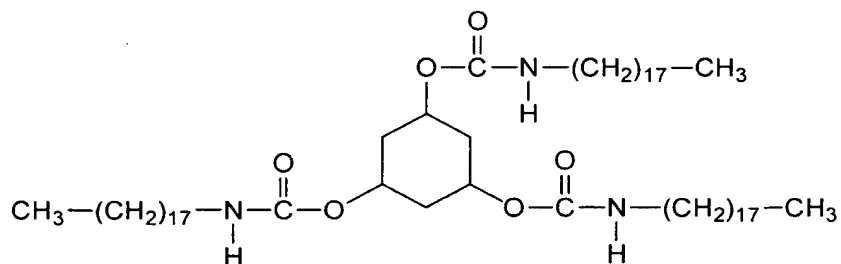
$\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by 3 parts by weight of the following compound.



5

#### Example 20

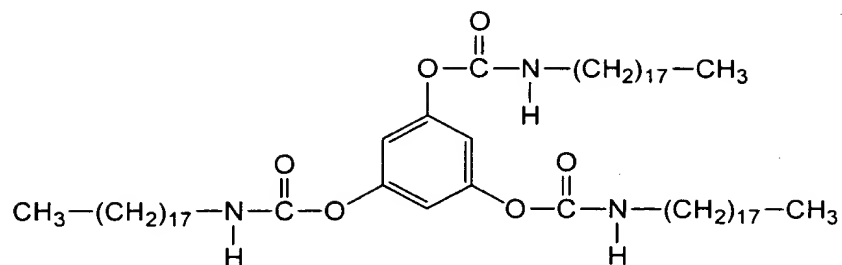
A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and  
 10  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by 3 parts by weight of the following compound.



#### Example 21

15 A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 7 parts and  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCOCONH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by 3 parts by weight

of the following compound.



5 Example 22

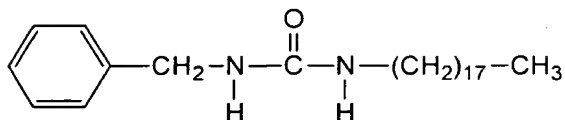
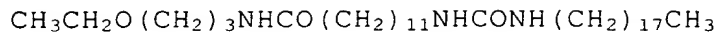
A thermo reversible recording medium was produced in the same manner as in Example 16 except that behenyl behenate was replaced by ethanolamine distearate (manufactured by Nippon Kasei Chemical Co., Ltd., Suliaid S).

10 Example 23

A thermo reversible recording medium was produced in the same manner as in Example 16 except that behenyl behenate was replaced by diheptadecylketone (manufactured by Nippon Kasei Chemical Co., Ltd., Wax KS).

15 Example 24

A thermo reversible recording medium was produced in the same manner as in Example 1 except that the amount of behenyl behenate was changed to 9 parts and  $\text{CH}_3(\text{CH}_2)_{17}\text{NHCOCNH}(\text{CH}_2)_{17}\text{CH}_3$  was replaced by each 0.5 parts  
20 of the following two compounds.



[Evaluation]

5           The following evaluations were conducted using thus produced thermo reversible recording media (Examples 1 to 13, 16 to 24, and Comparative Examples 1 and 2).

(1) Contrast

10           The medium obtained as described above was ~~cleared~~ made transparent previously, and ~~heat was applied on this~~ heated using a magnetic card readerwriter having print deleting function (KU-R-3001FA) manufacture by Kyushu Matsushita Electric Co., Ltd., while setting the printing energy value of a thermal heat to 0.47 mJ/dot, the medium

15           was cooled to room temperature, and the reflection ~~concentration~~ density ("initial image density") was measured ~~by using a Macbeth reflection densitometer to obtain the initial image concentration.~~

20           Then, opaque images were formed on the medium using the same apparatus at the above-mentioned energy value, thermal printing was conducted while ~~wetting~~ setting the deletion temperature value of a ceramic heater to 90 °C, the medium was cooled to room temperature, and the reflection ~~concentration~~ density ("initial deletion density") was

measured ~~by using a~~ the Macbeth reflection densitometer ~~to~~  
obtain the initial deletion concentration.

Next, the initial contrast was calculated ~~from the~~  
~~above obtained initial image concentration and the initial~~  
5 ~~deletion concentration (~~ as initial deletion concentration  
~~- initial image concentration)~~.

(2) Ammonia resistance

A previously ~~cleared~~ transparent made medium was  
immersed in a 8% ammonia aqueous solution for 48 hours, then,  
10 the initial image concentration density and initial deletion  
~~concentration density~~ were measured ~~by using~~ the same  
apparatus according to the same method as used in the  
above-mentioned contrast measurement, and these were called  
the image ~~concentration density~~ after test, and the deletion  
15 ~~concentration density~~ after the test.

~~Then, the~~ The contrast after the test was calculated  
~~from the above obtained image concentration after test and~~  
~~deletion concentration after test (~~ as deletion  
~~concentration density after the test - image concentration~~  
20 density after the test).

(3) Deleting property

[A] The medium was previously ~~cleared~~ made transparent,  
and partially opacified using a readerwriter (KU-R-3001FA)  
manufacture by Kyushu Matsushita Electric Co., Ltd., under  
25 5 °C by a thermal inclination tester, then, deletion was

effected at the optimum deletion temperature around the center of the deletion temperature. This procedure was conducted on 50 pieces of paper, and deleted condition of images was visually confirmed and evaluated.

5 Evaluation level:

O: all can be deleted

O to Δ: pale deletion residue occurs slightly

Δ: pale deletion residue occurs sometimes

x: pale deletion residue often occurs

10 The results of (1) to (3) are summarized in Table 1.

(4) ~~Clearing~~ Transparent-state upper limit temperature (~~T<sub>tu</sub>~~ T<sub>wx</sub>), ~~opacification~~ opaque-state lower limit temperature (~~T<sub>sl</sub>~~ T<sub>vo</sub>), temperature difference ( ~~ΔT<sub>ts</sub>~~ ΔT<sub>wv</sub>) between ~~clearing~~ transparent-state upper limit temperature and ~~opacification~~ opaque-state lower limit temperature, ~~clearing~~ transparent-state lower limit temperature (~~T<sub>tl</sub>~~ T<sub>wo</sub>), ~~clearing~~ and transparent-state temperature range ( ΔT<sub>w</sub> ΔT<sub>z</sub>)

20 ~~T<sub>tu</sub>, T<sub>sl</sub>, ΔT<sub>ts</sub>, T<sub>tl</sub> and ΔT<sub>w</sub>~~ T<sub>wx</sub>, T<sub>vo</sub>, ΔT<sub>wv</sub>, T<sub>wo</sub>, ΔT<sub>z</sub> were measured using the above-mentioned measuring methods. The results are summarized in Tables 3 and 4.

Table 1

	Initial			Ammonia resistance test			Deleting property
	Image density	Deletion density	Contrast	Image density	Deletion density	Contrast	
Example 1	0.22	1.24	1.02	0.23	1.25	1.02	O
Example 2	0.20	1.20	1.00	0.22	1.23	1.01	O
Example 3	0.19	1.14	0.95	0.23	1.15	0.92	O~Δ
Example 4	0.18	1.12	0.94	0.20	1.14	0.94	Δ
Example 5	0.19	1.20	1.01	0.21	1.25	1.04	O
Example 6	0.18	1.18	1.00	0.21	1.23	1.02	O
Example 7	0.18	1.17	0.99	0.19	1.20	1.01	O
Example 8	0.19	1.16	0.97	0.22	1.19	0.97	O
Example 9	0.20	1.18	0.98	0.23	1.24	1.01	O
Example 10	0.18	1.17	0.99	0.20	1.22	1.02	O
Example 11	0.19	1.22	1.03	0.20	1.25	1.05	O
Example 12	0.18	1.20	1.02	0.21	1.22	1.01	O
Example 13	0.22	1.24	1.02	0.25	1.26	1.01	Δ
Comparative Example 1	0.24	1.10	0.86	1.08	1.11	0.03	Δ
Comparative Example 2	0.30	0.98	0.68	0.34	1.00	0.66	×



Table 2

	Initial			Ammonia resistance test			Deleting property
	<u>Image density</u>	<u>Deletion density</u>	<u>Contrast</u>	<u>Image density</u>	<u>Deletion density</u>	<u>Contrast</u>	
Example 16	0.18	1.19	1.01	0.20	1.20	1.00	O
Example 17	0.18	1.20	1.02	0.20	1.21	1.01	O
Example 18	0.22	1.15	0.93	0.24	1.16	0.92	O
Example 19	0.21	1.14	0.93	0.24	1.15	0.91	O
Example 20	0.18	1.22	1.04	0.21	1.23	1.02	O
Example 21	0.23	1.25	1.02	0.25	1.26	1.01	O
Example 22	0.23	1.23	1.00	0.25	1.25	1.00	O
Example 23	0.22	1.22	1.00	0.24	1.23	0.99	O
Example 24	0.18	1.25	1.07	0.21	1.27	1.06	O

Table 3

	Trans- parent- state upper limit temper- ature ( $T_{tu}$ $T_{wx}$ )	Opaque- state lower limit temper- ature ( $T_{sl}$ $T_{vo}$ )	Temper- ature differ- ence ( $\Delta T_{ts}$ $\Delta T_{wv}$ )	Trans- parent- state lower limit temper- ature ( $T_{tl}$ $T_{wo}$ )	Trans- parent- state temper- ature range ( $\Delta T_w$ $\Delta T_z$ )
Example 1	125	136	11	78	47
Example 2	120	127	7	76	44
Example 3	116	133	7	76	30
Example 4	110	116	6	75	25
Example 5	121	128	7	80	41
Example 6	123	133	10	82	41
Example 7	135	144	9	83	53
Example 8	133	146	13	85	48
Example 9	132	140	8	84	48
Example 10	115	123	8	81	34
Example 11	120	130	10	83	37
Example 12	140	148	8	85	55
Example 13	114	126	12	87	27
Compar- ative Example 1	123	133	9	98	25
Compar- ative Example 2	104	125	21	81	23

Table 4

	Trans- parent- state upper limit temper- ature ( $T_{tu}$ $T_{wx}$ )	Opaque- state lower limit temper- ature ( $T_{sl}$ $T_{vo}$ )	Temper- ature differ- ence ( $\Delta T_{ts}$ $\Delta T_{wv}$ )	Trans- parent- state lower limit temper- ature ( $T_{tl}$ $T_{wo}$ )	Trans- parent- state temper- ature range ( $\Delta T_w$ $\Delta T_z$ )
Example 16	121	127	6	80	41
Example 17	122	130	8	77	45
Example 18	125	132	7	79	46
Example 19	134	140	6	78	56
Example 20	115	119	4	78	38
Example 21	123	130	7	79	45
Example 22	116	123	7	83	33
Example 23	118	124	6	82	36
Example 24	128	136	8	80	48

According to the present invention, a thermo reversible recording medium which can give sufficient degree of opacification even if preserved in the presence of a basic substance, ~~expands-clearing~~ has wider transparent-state temperature range, can provide an image having sufficient image deleting property and high contrast even if the environment temperature changes, and can provide sufficient degree of opacification, a member having ~~an information memorizing part~~ a memory, an image processing method, and image processing apparatus are provided. Thus, ~~contribution~~ this invention has considerably contributed to the field of the thermo reversible recording ~~field is~~ extremely large.

The present document incorporates by reference the entire contents of Japanese priority documents, 2000-176727 filed in Japan on June 13, 2000.

Although the invention has been described with respect to a specific embodiment for a complete and clear disclosure, the appended claims are not to be thus limited but are to be construed as embodying all modifications and alternative constructions that may occur to one skilled in the art which fairly fall within the basic teaching herein set forth.

ABSTRACT OF THE DISCLOSURE

The thermo reversible recording medium comprises a substrate and a ~~heat~~ thermo sensible layer. This ~~heat~~ thermo sensible layer is made of resin and organic lower molecular weight substance and can ~~becoming transparent or non-transparent or vice versa~~ become transparent-state or opaque-state depending on temperature. The organic lower molecular weight substance is a linear hydrocarbon-containing compound having no carboxyl group (A) and a linear hydrocarbon-containing compound having no carboxyl group (B) having a melting point lower than the melting point of the linear hydrocarbon-containing compound having no carboxyl group (A) by 20 °C or more.

**IN THE UNITED STATES PATENT AND TRADEMARK OFFICE**

Applicants: Yoshihiko HOTTA et al.

Serial No.: 09/877,140

Group Art Unit 1774

Filed : June 8, 2001

Examiner Bruce H. Hess

For : THERMO REVERSIBLE RECORDING MEMBER  
HAVING INFORMATION MEMORIZING PART,  
THERMO REVERSIBLE RECORDING LABEL,  
METHOD OF AND APPARATUS FOR IMAGE  
PROCESSING

**ANNOTATED MARKED-UP DRAWINGS**

**(SHEET CONTAINING FIGS. 1 AND 2)**

[illegible]